

## Periodic Trends Worksheet

*Directions: Use your notes to answer the following questions.*

1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Oxygen < Carbon < Aluminum < Potassium

2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Neon < Aluminum < Sulfur < Oxygen

3. Why does fluorine have a higher ionization energy than iodine?

Fluorine has nine protons attracting 9 electrons which are much closer to the nucleus compared to iodine. This results in smaller size of Fluorine than iodine making it difficult to remove an electron from the atom.

4. Why do elements in the same family generally have similar properties?

Because they have same number of electrons in the outer shell (valence electrons) which take part in chemical reaction.

5. Indicate whether the following properties increase or decrease from left to right across the periodic table.

- |  |           |
|--|-----------|
| a. atomic radius (excluding noble gases) | Decreases |
| b. first ionization energy               | Increases |
| c. electronegativity                     | Increases |

6. What trend in atomic radius occurs down a group on the periodic table? What causes this trend?

Atomic radius **decreases** down the group on the periodic table. As we compare the elements down the group, the effective nuclear charge increases, but at the same time the outermost electrons are found in the shell that is farther away from the nucleus. Also the number of screening electrons increases. This results in reduced attraction between the nucleus and the outermost electrons.

7. What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ionization energy **increases** from left to right across the period. The atomic size becomes smaller from left to right. So it becomes harder to remove electron from the atom. Hence the energy required to do so (ionization Energy) increases.

# Periodic Trends Reactivity Lab Answer Key

**Michael Brown**



## **Periodic Trends Reactivity Lab Answer Key:**

This book delves into Periodic Trends Reactivity Lab Answer Key. Periodic Trends Reactivity Lab Answer Key is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Periodic Trends Reactivity Lab Answer Key, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Periodic Trends Reactivity Lab Answer Key
- Chapter 2: Essential Elements of Periodic Trends Reactivity Lab Answer Key
- Chapter 3: Periodic Trends Reactivity Lab Answer Key in Everyday Life
- Chapter 4: Periodic Trends Reactivity Lab Answer Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Periodic Trends Reactivity Lab Answer Key. The first chapter will explore what Periodic Trends Reactivity Lab Answer Key is, why Periodic Trends Reactivity Lab Answer Key is vital, and how to effectively learn about Periodic Trends Reactivity Lab Answer Key.

3. In chapter 2, this book will delve into the foundational concepts of Periodic Trends Reactivity Lab Answer Key. This chapter will elucidate the essential principles that must be understood to grasp Periodic Trends Reactivity Lab Answer Key in its entirety.

4. In chapter 3, the author will examine the practical applications of Periodic Trends Reactivity Lab Answer Key in daily life. The third chapter will showcase real-world examples of how Periodic Trends Reactivity Lab Answer Key can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Periodic Trends Reactivity Lab Answer Key in specific contexts. This chapter will explore how Periodic Trends Reactivity Lab Answer Key is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Periodic Trends Reactivity Lab Answer Key. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Periodic Trends Reactivity Lab Answer Key.

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