

Periodic Trends

ATOMIC RADIUS

1. What trend in atomic radius do you see as you go down a group/family on the periodic table?
2. What causes this trend?
3. What trend in atomic radius do you see as you go across a period/row on the periodic table?
4. What causes this trend?
5. Circle the atom in each pair that has the largest atomic radius.
 - a) Al B
 - b) S O
 - c) Br Cl
 - d) Na Al
 - e) O F
 - f) Mg Ca

6. Put the following elements in order from smallest to largest atomic radius **and** explain why:
C, O, Sn, Sr.

ELECTRONEGATIVITY

7. Define electronegativity
8. How does the ionic radius of a nonmetal compare with its atomic radius?
9. What trend in electronegativity do you see as you go down a group/family on the periodic table?
10. What causes this trend?
11. What trend in electronegativity do you see as you go across a period/row on the periodic table?
12. What causes this trend?
13. Circle the atom in each pair that has the greater electronegativity.
 - a) Ca Ga
 - b) Li O
 - c) Cl S
 - d) Br As
 - e) Ba Sr
 - f) O S

GENERAL QUESTIONS

14. Which group tends to form +1 ions? _____
15. Which group tends to form +2 ions? _____
16. Which group tends to form -1 ions? _____
17. Which group tends not to form ions or react? _____

Periodic Trends Answer Key

Hongru Du

A red circular graphic with a gradient, appearing as a stylized arrow or a decorative element, located to the right of the author's name.

Periodic Trends Answer Key:

Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Periodic Trends Answer Key**

In a global inundated with screens and the cacophony of immediate conversation, the profound energy and emotional resonance of verbal beauty usually disappear in to obscurity, eclipsed by the regular barrage of noise and distractions. Yet, nestled within the musical pages of **Periodic Trends Answer Key**, a charming function of literary splendor that pulses with raw feelings, lies an wonderful trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, that magical opus courses readers on an emotional odyssey, gently revealing the latent potential and profound impact stuck within the intricate internet of language. Within the heart-wrenching expanse of the evocative examination, we can embark upon an introspective exploration of the book is key styles, dissect their fascinating publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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