

## Periodic Table Named-Groups Worksheet – 2

**DIRECTIONS:** Use the Periodic Table shown below to answer each question, using the letters in each "box" to identify the required element(s).

A																B
											D		E			
C																
	F					G					H					K
	L												I			
J		Z			M						N			O		P
Q		R			S	T	U									

  

V						W										
X						Y										

List elements in alphabetical order when more than 1 element answers a question.

- \_\_\_\_\_ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas   **B)** it is a gas   **C)** it is NOT a gas   **D)** it does NOT share the properties of the other elements in its group   **E)** it is a liquid
- \_\_\_\_\_ 2) Which of these elements is/are highly reactive? **A)** "C"   **B)** "Z"   **C)** "K"   **D)** A, B, and C   **E)** A and B only
- \_\_\_\_\_ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- \_\_\_\_\_ 4) Which letter labels the halogen element with the lowest atomic number?
- \_\_\_\_\_ 5) Which of these elements is the least reactive? **A)** "J"   **B)** "Y"   **C)** "A"   **D)** "O"   **E)** "B"
- \_\_\_\_\_ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C"   **B)** "N"   **C)** "M"   **D)** "W"   **E)** "Y"

continued on other side

# Periodic Table Worksheet Contestado

**Marcel A. Müller**



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