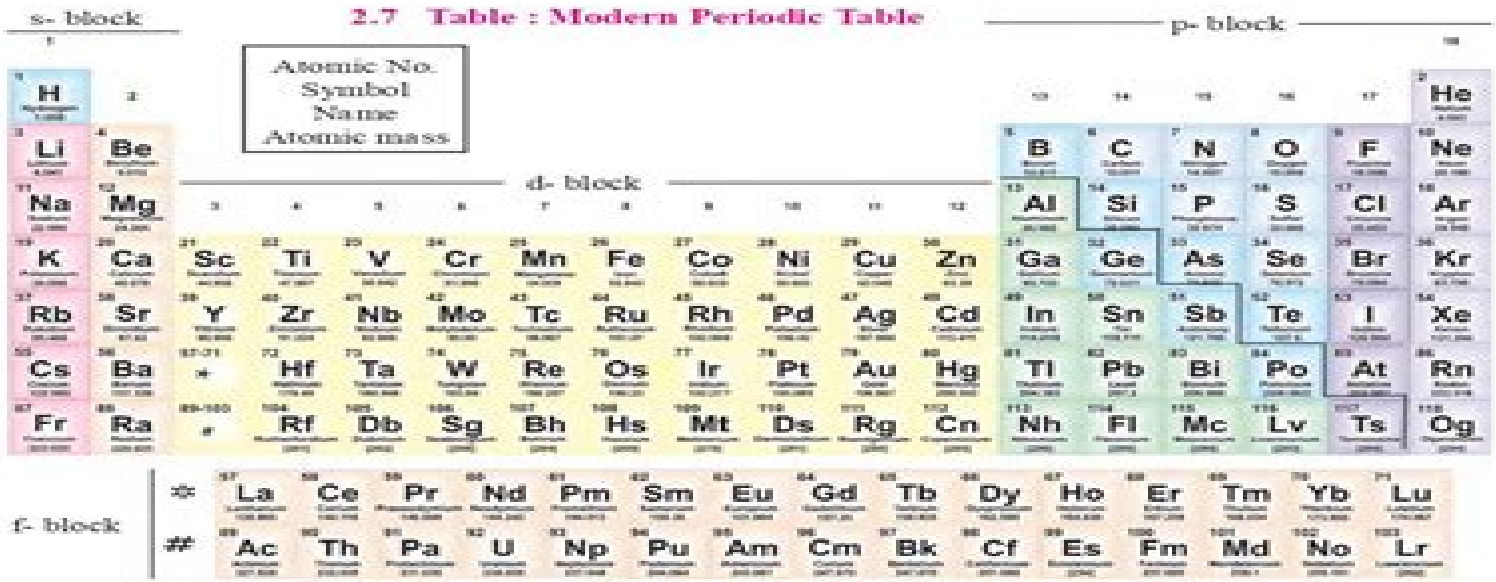
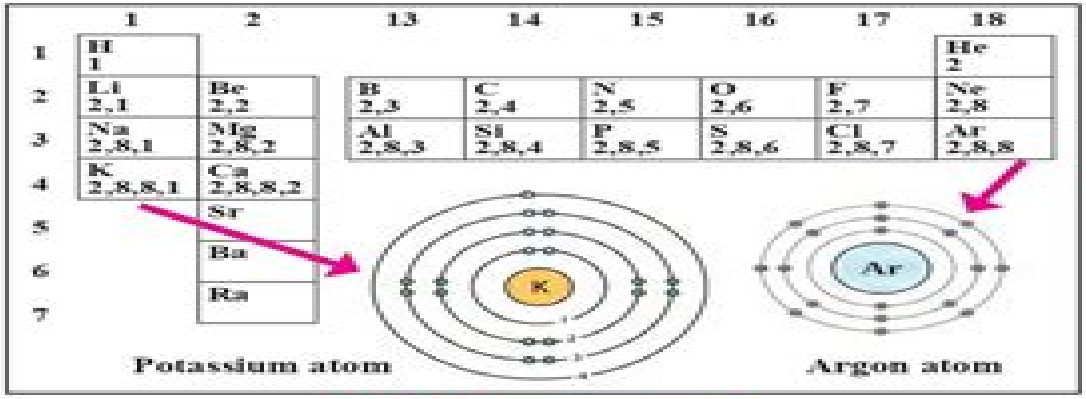
2.7 Table : Modern Periodic Table



Periods and electronic configuration

1. On going through the modern periodic table it is seen that the elements Li, Be, B, C, N, O, F and Ne belong to the period-2. Write down electronic configuration of all of them.
2. Is the number of valence electrons same for all these elements?
3. Is the number of shells the same in these ?

You will find that the number of valence electrons is different in these elements. However, the number of shells is the same. You will also find that, while going from left to right, within the period, the atomic number increases by one at a time and the number of valence electrons also increases by one at a time.



2.8 New period new shell

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1. Understanding the eBook Periodic Table Physical Science If8767 Answers
    - The Rise of Digital Reading Periodic Table Physical Science If8767 Answers
    - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Physical Science If8767 Answers
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Periodic Table Physical Science If8767 Answers
    - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Physical Science If8767 Answers
    - Personalized Recommendations
    - Periodic Table Physical Science If8767 Answers User Reviews and Ratings
    - Periodic Table Physical Science If8767 Answers and Bestseller Lists

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    - Periodic Table Physical Science If8767 Answers Public Domain eBooks
    - Periodic Table Physical Science If8767 Answers eBook Subscription Services
    - Periodic Table Physical Science If8767 Answers Budget-Friendly Options
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    - ePub, PDF, MOBI, and More
    - Periodic Table Physical Science If8767 Answers Compatibility with Devices
    - Periodic Table Physical Science If8767 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Periodic Table Physical Science If8767 Answers
    - Highlighting and Note-Taking Periodic Table Physical Science If8767 Answers
    - Interactive Elements Periodic Table Physical Science If8767 Answers
8. Staying Engaged with Periodic Table Physical Science If8767 Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Periodic Table Physical Science If8767 Answers
9. Balancing eBooks and Physical Books Periodic Table Physical Science If8767 Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Periodic Table Physical Science If8767 Answers
10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Physical Science If8767 Answers
    - Setting Reading Goals Periodic Table Physical Science If8767 Answers
    - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Physical Science If8767 Answers
    - Fact-Checking eBook Content of Periodic Table Physical Science If8767 Answers
    - Distinguishing Credible Sources
13. Promoting Lifelong Learning

    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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