

Post-Lab:

Research crystals and list other information you discovered. Use additional sheet of paper if needed but be sure to staple it to the lab sheets.

1. Where did rock candy originate?

Rock candy is one of the oldest and purest forms of candy. It was originally used by pharmacists to make medicines for many kinds of illnesses.

2. What is the name of crystals that you grew?

In the field of crystallography, these are called *monoclinic* crystals. Their shape is determined by the way the individual sugar molecules fit together, which is similar to the way the shape of a pile of oranges is determined by the shape of the individual oranges and the way they stack together.

3. What made the crystals grow?

As the water evaporates, sugar crystals form on the string or stick, and the shapes that they form reflect the shape of individual sugar crystals.

4. What was the purpose of soaking the skewer or string?

The string will provide the surface on which the crystals will grow. As water evaporates from the string, small crystals of sugar will encrust the string. These tiny seed crystals provide starting points for larger crystals. Future growth will be concentrated around these points.

5. What is a super saturated solution?

When you mixed the water and sugar you made a Super Saturated Solution which means that the water could only hold the sugar if both were very hot. As the water cooled, the sugar came out of the solution and went back into sugar crystals on the skewer.

Rock Candy Lab Chemistry Answers

TD Snyder



Rock Candy Lab Chemistry Answers:

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