

## Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

OH  
 NO<sub>3</sub>  
 C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>

hydroxide  
 nitrate  
 acetate

ClO<sub>3</sub>  
 HCO<sub>3</sub>  
 SO<sub>4</sub>  
 CO<sub>3</sub>

chlorate  
 bicarbonate  
 sulfate  
 carbonate

SO<sub>3</sub>  
 PO<sub>4</sub>  
 NH<sub>4</sub>

sulfite  
 phosphate  
 ammonium

- |  |   |
|--|---|
| 1. CuSO <sub>4</sub> _____                             | 26. NaOH _____  |
| 2. KClO <sub>3</sub> _____                             | 27. NaNO <sub>3</sub> _____                               |
| 3. NH <sub>4</sub> OH _____                            | 28. Ba(OH) <sub>2</sub> _____                             |
| 4. K <sub>2</sub> CO <sub>3</sub> _____                | 29. K <sub>2</sub> CO <sub>3</sub> _____                  |
| 5. Na <sub>2</sub> SO <sub>4</sub> _____               | 30. LiOH _____  |
| 6. KC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> _____ | 31. KHCO <sub>3</sub> _____                               |
| 7. NH <sub>4</sub> Cl _____                            | 32. NH <sub>4</sub> F _____                               |
| 8. Mg(OH) <sub>2</sub> _____                           | 33. KClO <sub>3</sub> _____                               |
| 9. Zn(NO <sub>3</sub> ) <sub>2</sub> _____             | 34. NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> _____  |
| 10. MgSO <sub>4</sub> _____                            | 35. CaClO <sub>3</sub> _____                              |
| 11. CaCO <sub>3</sub> _____                            | 36. SrNO <sub>3</sub> _____                               |
| 12. KOH _____  | 37. Be(OH) <sub>2</sub> _____                             |
| 13. Ca(OH) <sub>2</sub> _____                          | 38. Li <sub>2</sub> SO <sub>4</sub> _____                 |
| 14. K <sub>2</sub> CO <sub>3</sub> _____               | 39. Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> _____ |
| 15. AlPO <sub>4</sub> _____                            | 40. BaSO <sub>3</sub> _____                               |
| 16. Na <sub>2</sub> PO <sub>4</sub> _____              | 41. NH <sub>4</sub> I _____                               |
| 17. MgSO <sub>4</sub> _____                            | 42. NH <sub>4</sub> HCO <sub>3</sub> _____                |
| 18. AgNO <sub>3</sub> _____                            | 43. Cu <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub> _____ |
| 19. Na <sub>2</sub> CO <sub>3</sub> _____              | 44. Ni(OH) <sub>2</sub> _____                             |
| 20. NaHCO <sub>3</sub> _____                           | 45. ZnSO <sub>4</sub> _____                               |
| 21. Ca(NO <sub>3</sub> ) <sub>2</sub> _____            | 46. MnCO <sub>3</sub> _____                               |
| 22. BaSO <sub>4</sub> _____                            | 47. H <sub>2</sub> SO <sub>4</sub> _____                  |
| 23. BPO <sub>3</sub> _____                             | 48. Au <sub>2</sub> SO <sub>4</sub> _____                 |
| 24. LiNO <sub>3</sub> _____                            | 49. NH <sub>4</sub> NO <sub>3</sub> _____                 |
| 25. Al(NO <sub>3</sub> ) <sub>3</sub> _____            | 50. LiClO <sub>3</sub> _____                              |

# Polyatomic Ions Naming Answer

**Effiong Eyo**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

## **Polyatomic Ions Naming Answer:**

The book delves into Polyatomic Ions Naming Answer. Polyatomic Ions Naming Answer is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Polyatomic Ions Naming Answer, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Polyatomic Ions Naming Answer
- Chapter 2: Essential Elements of Polyatomic Ions Naming Answer
- Chapter 3: Polyatomic Ions Naming Answer in Everyday Life
- Chapter 4: Polyatomic Ions Naming Answer in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Polyatomic Ions Naming Answer. This chapter will explore what Polyatomic Ions Naming Answer is, why Polyatomic Ions Naming Answer is vital, and how to effectively learn about Polyatomic Ions Naming Answer.
3. In chapter 2, the author will delve into the foundational concepts of Polyatomic Ions Naming Answer. This chapter will elucidate the essential principles that must be understood to grasp Polyatomic Ions Naming Answer in its entirety.
4. In chapter 3, the author will examine the practical applications of Polyatomic Ions Naming Answer in daily life. The third chapter will showcase real-world examples of how Polyatomic Ions Naming Answer can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Polyatomic Ions Naming Answer in specific contexts. The fourth chapter will explore how Polyatomic Ions Naming Answer is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Polyatomic Ions Naming Answer. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Polyatomic Ions Naming Answer.

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