

# **FORMULAS WITH POLYATOMIC IONS**

Name \_\_\_\_\_

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions. The first one is done for you.

	$\text{OH}^-$	$\text{NO}_3^-$	$\text{CO}_3^{2-}$	$\text{SO}_4^{2-}$	$\text{PO}_4^{3-}$
$\text{H}^+$	$\text{HOH}$ $(\text{H}_2\text{O})$	$\text{HNO}_3$	$\text{H}_2\text{CO}_3$	$\text{H}_2\text{SO}_4$	$\text{H}_3\text{PO}_4$
$\text{Na}^+$					
$\text{Mg}^{+2}$					
$\text{NH}_4^+$					
$\text{Ca}^{+2}$					
$\text{K}^+$					
$\text{Al}^{+3}$					
$\text{Pb}^{+4}$					

# Polyatomic Ions Worksheet Answers Pogil

**LP Steffe**



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*Polyatomic Ions for Surface Analysis and Modification* Erick Ryan Fuoco, 2003      **The Particulate Nature of Polyatomic Ions**, 2007      *Unimolecular and Collision-induced Dissociation Study of Polyatomic Ions at High Collision Energy* Xuedong Zhou, 2001      *Standard Thermodynamic Functions of Gaseous Polyatomic Ions at 100-1000 K* Aharon Loewenschuss, Y. Marcus, 1987      **Structure of Free Polyatomic Molecules** K. Kuchitsu, Kozo Kuchitsu, 1998-09-10 This volume Structure of Free Polyatomic Molecules Basic Data contains frequently used data from the corresponding larger Landolt B rNSTEIN handbooks in a low price book for the individual scientists working in the laboratory Directories link to the more complete volumes in the library The book contains important information about a large number of semiconductors

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