TYESC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

1	Which of the following is a uni-univalent electrolyte?
	a) KCI
	b) C _i Cl ₂
	d) AlCli
	d) MgCl ₂
2	The ionic strength of a solution of molality m, and valency of the ion Zi is
	a) $\mu = \frac{\epsilon}{2} m_i Z \ell^2$
	b) $\mu = \sum_{i=1}^{n} m_i Z i^2$
	$\epsilon \lambda = \frac{1}{2} \sum Z \ell^2$
	d) $\mu = \frac{1}{2} \sum m_i Z \ell^2$
3	Which of the following is an example of electrolyte concentration cell with transference?
	ii) Cii [CuSO ₁] [AgNO ₁] Ag
	b) Pt. H _{2g} HCl _{sq} AgCl ₃₀ Ag
	c) Pt. H ₂₀ HCl _m (a) AgNO ₂ (a) Ag
	d) Ag – AgCl [NaCl(a ₀)] NaCl(a ₀)[AgCl - Ag
4	A concentration cell which has salt bridge in its representation is
	electrolyte concentration cell with transference
	b) Electrode concentration cell
	c) Electrolyte concentration cell without transference
	d) Gas concentration cell
5	The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by
	a) $E_j = E_{exidation} + E_{reduction}$
	b) $E_{cell} = E_{saldation} + E_{reduction} + E_{j}$
	c) $E_{cett} = E_{exidaction} - E_{reduction} - E_{j}$
	d) $E_j = E_{oxidation} - E_{reduction}$
6	The activity 'a' is related to the activity coefficient " y' by the equation
	a) a = = =
	b) $y = a - m$
	c) $\alpha = m + \gamma$
	d) $a = my$

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