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Statistical Mechanics Donald A. McQuarrie, 2000-06-16 The canonical ensemble Other ensembles and fluctuations Boltzmann statistics fermi dirac statistics and bose einstein statistics Ideal monatomic gas Ideal diatomic Classical statistical mechanics Ideal polyatomic Chemical equilibrium Quantum statistics Crystals Imperfect gases Distribution functions in classical monatomic liquids Perturbation theories of liquids Solutions of strong electrolytes Kinetic theory of gases and molecular collisions Continuum mechanics Kinetic theory of gases and the boltzmann equation Transport processes in dilute gases Theory of brownian motion The time correlation function formalism **Basic Physical Chemistry** E Brian Smith, 2012-06-26 This elegant book provides a student friendly introduction to the subject of physical chemistry It is concise and more compact than standard textbooks on the subject and it emphasises the two important concepts underpinning physical chemistry quantum mechanics and the second law of thermodynamics The principles are challenging to students because they both focus on uncertainty and probability The book explains these fundamental concepts clearly and shows how they offer the key to understanding the wide range of chemical phenomena including atomic and molecular spectra the structure and properties of solids liquids and gases chemical equilibrium and the rates of chemical reactions **Molecular Thermodynamics** Donald A. McQuarrie, John D. Simon, 1999-02-24 This text covers the principles of quantum mechanics before engaging those ideas in the subsequent development of thermodynamics It includes worked examples and chapter ending problems with solutions *Concise Physical Chemistry* Donald W. Rogers, 2011-03-31 This book is a physical chemistry textbook that presents the essentials of physical chemistry as a logical sequence from its most modest beginning to contemporary research topics Many books currently on the market focus on the problem sets with a cursory treatment of the conceptual background and theoretical material whereas this book is concerned only with the conceptual development of the subject Comprised of 19 chapters the book will address ideal gas laws real gases the thermodynamics of simple systems thermochemistry entropy and the second law the Gibbs free energy equilibrium statistical approaches to thermodynamics the phase rule chemical kinetics liquids and solids solution chemistry conductivity electrochemical cells atomic theory wave

mechanics of simple systems molecular orbital theory experimental determination of molecular structure and photochemistry and the theory of chemical kinetics Physical Chemistry Henry Eyring, Douglas Henderson, Wilhelm Jost, 1975

Mathematical Methods for Molecular Science John E. Straub, Joy Andrews, 2022-08-02 Straub's stunning new text is an excellent choice for a one semester course on mathematical methods an affordable supplement for physical chemistry courses or a self study guide This brilliant new text by John Straub Boston University is designed to bridge the mathematics knowledge gap between what is commonly known by students after completing a year of introductory calculus and what is required for success in the physical sciences and in physical chemistry courses Key concepts from the introductory calculus sequence are reviewed and carefully selected topics in multivariate calculus probability and statistics ordinary differential equations and linear algebra are explored Additional chapters cover advanced topics including partial differential equations Fourier analysis and group theory Engaging narratives fully worked examples hundreds of colorful visualizations and ample end of chapter problems with complete answers combine to make this stunning new text an excellent choice for a one semester course on mathematical methods as a supplement for courses in physical chemistry or as a self study guide

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well with McQuarrie Simon's landmark P CHEM text but can be used with others as well **A Conceptual Guide to Thermodynamics** Bill Poirier, 2014-07-16 Thermodynamics is the science that describes the behavior of matter at the macroscopic scale and how this arises from individual molecules. As such, it is a subject of profound practical and fundamental importance to many science and engineering fields. Despite extremely varied applications ranging from nanomotors to cosmology, the core concepts of thermodynamics such as equilibrium and entropy are the same across all disciplines. A Conceptual Guide to Thermodynamics serves as a concise conceptual and practical supplement to the major thermodynamics textbooks used in various fields. Presenting clear explanations of the core concepts, the book aims to improve fundamental understanding of the material as well as homework and exam performance. Distinctive features include Terminology and Notation Key: A universal translator that addresses the myriad of conventions, terminologies, and notations found across the major thermodynamics texts; Content Maps: Specific references to each major thermodynamic text by section and page number for each new concept that is introduced; Helpful Hints and Don't Try Its: Numerous useful tips for solving problems as well as warnings of common student pitfalls; Unique Explanations: Conceptually clear, mathematically fair, simple yet also sufficiently precise and rigorous. A more extensive set of reference materials, including older and newer editions of the major textbooks as well as a number of less commonly used titles, is available online at <http://www.conceptualthermo.com>. Undergraduate and graduate students of chemistry, physics, engineering, geosciences, and biological sciences will benefit from this book, as will students preparing for graduate school entrance exams and MCATs.

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