

- a) Benzene.
- b) Octane.
- c) Pentane.
- d) Entanol.

ANSWER : Pentane.

39) When a solute is dissolved in water it shows:

- a) decrease in freezing point of water.
- b) decrease in boiling point of water.
- c) increase in vapour pressure of water.
- d) all of the above.

ANSWER : decrease in freezing point of water.

40) When sugar is added to water, what is the change observed in boiling and freezing points of water?

- a) Both boiling point and freezing point decreases.
- b) Both boiling point and freezing point increases.
- c) Boiling point increases and freezing point decreases.
- d) Boiling point decreases and freezing point increases.

ANSWER : Boiling point increases and freezing point decreases.

41) aqueous metal solution have highest freezing point.

- a) Urea
- b) Barium chloride.
- c) Potassium bromide.
- d) Aluminium sulphate.

ANSWER : Urea

42) When a thin slice of sugar beet is placed in a concentrated solution of NaCl then sugar beet will.....

- a) dissolve in solution.
- b) neither absorb nor lose water.
- c) absorb water from solution.
- d) lose water from its cell.

ANSWER : lose water from its cell.

43) The wrong relation between osmotic pressure (P), volume (V) and temperature (T) is:

- a) $P \propto n$ if T and V are constant.
- b) Pv is constant if T is constant.
- c) $P \propto V$ if T is constant.
- d) $P \propto T$ if V is constant.

ANSWER : $P \propto V$ if T is constant.

44) Two solutions C and D are separated by a semi-permeable membrane. If liquid flows from D to C then.

- a) Both have same concentration.
- b) D is less concentrated than C.
- c) D is more concentrated than C.
- d) None of these.

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amongst one the most prestigious medical entrance exams in India With just a few months left for the examination it becomes quite challenging for students to cover all the concepts included in the NEET syllabus thoroughly However a proper study plan designed as per the latest examination pattern and the syllabus can help students to prepare all the important concepts in shorter time duration Given below are few useful tips that can assist the students in tackling multiple choice questions in NEET exam accurately In most of the multiple choice questions the options are designed in a very tricky and confusing manner In most of the cases all the given options seem to be correct in some aspect Therefore the students are advised to read the entire question very carefully Try to accumulate all the information provided in the question effectively because in some of the cases you can easily evaluate the Answers from the question itself If you are muddled by the given options then give each option a true and false test Instead of getting confused consider all the possibilities and neglect the incorrect options Hence in this way the most appropriate answer could be easily spotted Topic wise MCQs for NEET Chemistry Environmental Chemistry Organic Chemistry P Block Elements States of Matter Redox Reaction Chemical Equilibrium Thermodynamics Periodic Classification of Elements s Block Elements Chemical Bonding Molecular Structure Structure of Atom Some Basic Concepts of Chemistry Hydrocarbons Some Basic Concepts of Chemistry Solid State Structure of Atom Solutions Classification of Elements and Periodicity in Properties Electrochemistry Chemical Bonding and Molecular Structure Chemical Kinetics States of Matter Gases and Liquids Surface Chemistry Thermodynamics General Principles and Processes of Isolation of Elements Equilibrium Redox Reactions d and f Block Elements Hydrogen Coordination Compounds Haloalkanes and Haloarenes Some p Block Elements Alcohols Phenols and Ethers Organic Chemistry Some Basic Principles and Techniques Aldehyde Ketones and Carboxylic Acids Hydrocarbons Organic Compounds Containing Nitrogen Environmental Chemistry Biomolecules Polymers and Chemistry in Everyday Life Use a step wise approach to solve conceptual and complex questions Several times Matching type Questions are asked where the students are required to find the mismatched or the correctly matched option Some of the questions asked in the NEET exam are entirely memory based therefore the students are advised to memorize the common names scientific names concepts and important definitions Around 40% of the questions asked in the NEET exam are application based Therefore students need to focus more on the concepts along with its applications in order to score well in the examination Chemistry topics broadly can be grouped into Organic Inorganic and Physical chemistry Organic Chemistry is a vast section and holds the maximum weightage Its concepts and principles are important One needs to be conversant with properties of groups conversions recognizing organic compounds etc Interlinking between different chapters from organic chemistry can be of great help The students must primarily focus on reading NCERT textbooks Several times the questions asked in NEET exam are taken directly from the NCERT textbooks Initially avoid answering those questions for which you are not confident because your wrong answer may reduce your final score In order to utilize your time appropriately divide the three hours of examination time as per your

comfort among Physics Chemistry and Biology Initially focus on attempting all easy questions and later on pick the difficult ones By this way your confidence will be elevated and you will also get more time to answer hard questions Practice previous years question papers mock tests and sample papers to get an idea on how to answer MCQ questions efficiently Preparing at an early stage is what an MCQ exam requires Avoid guesswork for negative marking questions as they might lower your final score These tips can be very helpful for students to answer difficult and brain teaser questions Prior preparations and practice are mandatory aspects of any examination Hence to crack highly competitive examination like NEET it is mandatory for students to prepare well and acquire the skills to tackle multiple choice questions effectively Rather than just following mere guesswork the aspirants can focus on the tips discussed to tackle Multiple Choice Questions in NEET in the right manner

Problems in Physical Chemistry JEE Main and Advanced Volume 1 Dr. RK Gupta, 2021-04-05 1 The book is prepared for the problem solving in chemistry 2 It is divided into 8 chapters 3 Each chapter is topically divided into quick theory Immediate Test and Knowledge Confirmation Test 4 At the end of the each chapter cumulative exercises for JEE Main Advanced for practice 5 Acid Test for JEE Mains Advance containing all types of questions asked in JEE A common phrase among JEE Aspirants that chemistry is the most scoring subject but the problems asked in JEE Exams are not directly related but they are based on multiple applications Introducing the all new edition of Problem Physical Chemistry JEE Main Advanced Volume 1 which is designed to develop the use of the concepts of chemistry in solving the diversified problems as asked in JEE The book divides the syllabus into 8 chapters and each chapter has been topically divided in quick theory different types of Solved Examination followed by Immediate Test along with the Topicwise short exercises Knowledge Confirmation Test At the end of each chapter there are separate cumulative exercises for JEE Main Advanced Acid Test for JEE Mains Advance are also provided containing all types of questions asked in JEE Detailed and explanatory solutions provided to all the questions for the better understanding TOC Mole concept and Stiochiometry Atomic Structure Stages of Matter 1 Stages of Matter 2 Thermodynamic Thermochemistry Chemical Equilibrium Ionic Equilibrium *A Textbook of Organic Chemistry, 4th Edition* Tewari, K.S. & Vishnoi, N.K., The book *A Textbook of Organic Chemistry* was first published 40 years ago Over the years it has become students favourite because it explains the subject in the most student friendly way and is revised regularly to keep itself updated with the latest in research This edition presents the modern day basic principles and concepts of the subject as per the CBCS of UGC guidelines Special emphasis has been laid on the mechanism and electronic interpretation of reactions of the various classes of compounds It provides a basic foundation of the subject so that based on these students are able to extrapolate predict and solve challenging problems New in this Edition A new chapter Energy in Biosystems explores the fundamentals of biochemical reactions involved in storage as well as continuous usage of energy in biosystems Structural theories like VB and MO hybridization and orbital pictures of resonance and hyperconjugation Woodward Fieser rules for calculating max and Norrish type I and II reactions of special photochemical C

C cleavage in the chapter on Electromagnetic Spectrum Polanyi Hammond postulates and Curtin Hammett principle along with several new mechanisms e g Favorskii Baeyer Villiger and Birch in Chapter 5 McMurry Wittig Stobbe Darzen in Chapter 19 Study of antibiotics antacids and antihistamines in the chapter on Chemotherapy Biodegradable and conducting plastics in the chapter on Synthetic Polymers and Plastics Benefits of Green Chemistry the latest trend for sustainable chemistry as Appendix II

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