

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a 10x10 grid with the following layout:

- Top Row:** A (left), B (right)
- Second Row:** C (left), D (right)
- Third Row:** E (left), F (right)
- Fourth Row:** G (left), H (right)
- Fifth Row:** I (center), J (left), K (right)
- Sixth Row:** L (left), M (right)
- Seventh Row:** N (left), O (right)
- Eighth Row:** P (left), Q (right)
- Ninth Row:** R (left), S (right)
- Tenth Row:** T (left), U (right)

Arrows point from the center cell (I) to the bottom-left and bottom-right quadrants, labeled X and Y respectively.

List elements in alphabetical order when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- _____ 4) Which letter labels the halogen element with the lowest atomic number?
- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- _____ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

Periodic Tables Answer Key

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