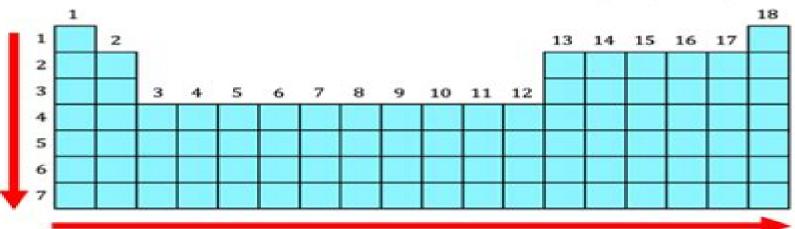
Periodic Trends - Atomic Mass

- The atomic mass describes the mass of each individual element.
- Atomic mass increases moving down each group and right across a period.

<u>Instructions</u>: In the periodic table below, draw two arrows representing the direction that atomic mass <u>increases</u> across groups and periods.



Instructions: In each question, identify the element with the greatest atomic mass:

1) Chlorine (Cl), Iodine (I), Bromine (Br), Fluorine (F)

Iodine

2) Oxygen (O), Carbon (C), Nitrogen (N), Boron (B)

Oxygen

3) Lithium (Li), Silicon (Si), Sulfur (Br), Xenon (Xe)

Xenon

Instructions: In each question, list the elements from least to greatest atomic mass:

4) Magnesium (Mg), Beryllium (Be), Barium (Ba)

Be, Mg, Ba

5) Gallium (Ga), Selenium (Se), Potassium (K)

K, Ga, Se

Periodic Table Trends Answer

L Darling-Hammond

Periodic Table Trends Answer:

Unveiling the Magic of Words: A Report on "Periodic Table Trends Answer"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really aweinspiring. Enter the realm of "**Periodic Table Trends Answer**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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