Name:	
Date:	Period:

#### Chemistry: The Periodic Table

Directions: Answer each of the following questions. You need not use complete sentences.

What is a period? How many are there in the periodic table?

A period is a row of elements on the periodic table, in which the elements are ordered one proton more each place from left to right. A period also often helps to represent an electron energy level and the order in which electrons are filled into their sublevels. There are 7 periods.

What is a group (also called a family)? How many are there in the periodic table?

Groups are the columns that are found top to bottom in the table. They show elements that have the same sublevel and number of electrons in their outer shell, valence electrons. Often elements in a group share any similar chemical properties because of this. There are 18 groups in the periodic table.

State the number of valence electrons in an atom of:

a. sulfur b. calcium c. chlorine d.arsenic 7 5

Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10 b. 18 c. 36 d. 90

Neon, Ne Argon, Ar Krypton, Kr Thorium, Th

List, by number, both the period and group of each of these elements.

	Symbol	Period	Group
a. beryllium	Be	2	2
b. Iron	Fe	-4	8
c. lead	Pb	6	1.4

Which of the following pairs of elements belong to the same period?

a. Na and Cl b. Na and Li c. Na and Cu d. Na and Ne

Which of the following pairs of elements belong to the same group?

a. H and He b. Li and Be c. C and Pb d. Ga and Ge

Would you expect strontium to be, chemically, more similar to calcium or rubidium and WHY?

Calcium, because they share a common amount of valence electrons. Valence electrons are the main contributor to chemical properties and how elements interact with other matter. Both strontium and calcium have two electron on the the "s" sublevel, 5s² and 4s² What are the Group 1 elements called?

# **Periodic Table Key Answers**

Tracy Budd, Heather Craven, Marc Davis, Mitchel Fedak, John P. Frade, Bernice Goldberg, Gary Land, Carol Rush

#### **Periodic Table Key Answers:**

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Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice Patterns In the Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives 

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