

# The Periodic Law

## SECTION 2

Select the correct answer for the following questions for the space provided.

1. \_\_\_\_\_ for the modern periodic table, elements are ordered
  - (a) according to increasing atomic mass
  - (b) according to Mendeleev's original design
  - (c) according to increasing atomic number
  - (d) based on how they were discovered
2. \_\_\_\_\_ Mendeleev's original short periodic table gave better chemical properties of elements appeared in regular intervals and whereas elements were arranged according to increasing
  - (a) density
  - (b) atomic number
  - (c) atomic mass
  - (d) electronegativity
3. \_\_\_\_\_ The modern periodic law states that
  - (a) all new elements with the same properties are listed in the same group
  - (b) the physical and chemical properties of elements are functions of atomic number
  - (c) elements exhibit properties of both periods and groups
  - (d) the chemical properties of elements can be grouped according to periodicity but physical properties cannot
4. \_\_\_\_\_ The discovery of the noble gases changed Mendeleev's periodic table
  - (a) adding a row
  - (b) adding a column
  - (c) adding a new group
  - (d) adding a new period
5. \_\_\_\_\_ The noble gases that property of the noble gases is that they are
  - (a) unstable
  - (b) inert
  - (c) radioactive
  - (d) highly reactive
6. \_\_\_\_\_ Likewise, the first column in Group 1, has characteristic number of 1. The second column in the group has characteristic number of
  - (a) 2
  - (b) 10
  - (c) 18
  - (d) 32
7. How groups in the periodic table are numbered? (a) and (b) atomic number of (a)
  - (a) 1 to 18
  - (b) 1 to 10
  - (c) 1 to 18
  - (d) 1 to 10
8. How many periods are there in the periodic table?
  - (a) 18
  - (b) 10
  - (c) 18
  - (d) 10
9. How many elements are in the periodic table?
  - (a) 118
  - (b) 110
  - (c) 118
  - (d) 110
10. What is the modern periodic table? How does it differ from the periodic table?
  - (a) 118
  - (b) 110
  - (c) 118
  - (d) 110

# Periodic Law Chapter 5 Test Answers

**G Psacharopoulos**



## **Periodic Law Chapter 5 Test Answers:**

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