



## Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H<sub>2</sub>O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

# Percent Yield Calculations Answers

**RJ Alexander**



## **Percent Yield Calculations Answers:**

## Embracing the Song of Phrase: An Mental Symphony within **Percent Yield Calculations Answers**

In a global consumed by screens and the ceaseless chatter of fast transmission, the melodic elegance and emotional symphony produced by the published term often disappear in to the back ground, eclipsed by the persistent noise and disruptions that permeate our lives. But, set within the pages of **Percent Yield Calculations Answers** a stunning literary value overflowing with raw thoughts, lies an immersive symphony waiting to be embraced. Crafted by a wonderful musician of language, that captivating masterpiece conducts readers on a psychological journey, skillfully unraveling the hidden melodies and profound affect resonating within each cautiously crafted phrase. Within the depths of the emotional analysis, we shall examine the book is central harmonies, analyze their enthralling publishing design, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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