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# Pastel Partner Training Manual

**Patrick Vollmar**



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web elements in the second row of the periodic table place their electrons in the  $2n$  shell as well as the  $1n$  shell for instance lithium li has three electrons two fill the  $1s$  orbital and the third is placed in the  $2s$  orbital giving an electron configuration

of 1 s 2 2 s 1

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium he 2s<sup>2</sup> 2 is an s electron whereas the electron removed during the ionization of boron he 2s<sup>2</sup> 2p<sup>1</sup> is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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