STD XII - CHEMISTRY

Time: 3 hrs.]

(a) 68%

MARCH - 2013

[Marks: 70

7

SECTION - I

Q. 1.	Select and write the most appropriate answers from the given
	alternatives for each sub-question:
(i)	In body centred cubic structure the enece occupied is about

(b) 53% (c) 38% (ii) For a gaseous reaction the unit of rate of reaction is

(d) 32%

(a) L atm s⁻¹ (b) atm mol-1 s-1

 $(c, stm s^{-1})$

(iii) Which of the following compounds contains S = O as well as S = S bonds? (a) Sulphuric acid (c) Sulphurous acid

(b) Thiosulphuric aci (d) Thiosulphurous acid

(iv) Which of the following solutions shows maximum depression in freezing point? (a) 0.5 M Li₂SO₄ (b) 1 M NaCl (c) 0.5 M Al₂(SO₄)₃ (d) 0.5 M BaCl₂

(v) For a chemical reaction dS = − 0.035 kJ/K and dH = −20 kJ. At what temperature does the reaction turn nonspontaneous? (a) 5.14 K (b) 57.14 K (c) 571.4 K (d) 5714.0 K

(vi) The standard e.m.f. of the following cell is 0.463 V

 $Cu_{(s)}$ $Cu_{(\alpha q)}^{++}$ $Ag_{(\alpha q)}^{++}$ $Ag_{(s)}$. What is the standard potential of Cu electrode?

(a) 1.137 V

(b) 0.337 V

(c) 0.463 V

(d) -0.463 V

(vii) Fe₂O₃ is reduced to spongy iron near the top of blast furnace by ____ (a) H. (b) CaO (c) SiO₂ (d) CO

Q. 2. Answer any SIX of the following:

[12]

(i) Distinguish between crystalline solid and amorphous solid.

(ii) State Kohlrausch Law and write mathematical expression of molar conductivity of the given solution at infinite dilution.

(iii) Write cell reactions in lead storage battery during discharge.

(iv) Draw structures and write geometry of PCl₃ and PCl₅.

(v) Prove that ΔH = ΔU + Δ nRT. What is the condition under which ΔU = ΔH?

(vi) Mention names and formulae of two ores of aluminium.

(vii) Derive the relationship between relative lowering of vapour pressure and molar mass of nonvolatile solute.

(viii) What is pseudo first order reaction? Give one example of it.

Q. 3. Answer any THREE of the following:

[9]

(i) Calculate the mole fraction and molality of HNO3 in a solution containing 12.2% HNO3. (Given - atomic masses: H = 1, N = 14, O = 16)

(ii) Consider the reaction —

$$3I_{(aq)}^{-} + S_2O_8^{2-} \longrightarrow I_{3(aq)}^{-} + 2SO_{4(aq)}^{2-}$$
 at particular time t, $\frac{d[SO_4^{2-}]}{dt} = 2.2 \times 10^{-2} \text{ M/s}.$

What are the values of the following at the same time?

$$(a) - \frac{d[I^-]}{dt}$$

(b)
$$-\frac{d\left[S_2O_8^{2-}\right]}{dt}$$

(c)
$$-\frac{d\left[I_3^-\right]}{dt}$$

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- 1. Understanding the eBook Past Hsc Exam Papers
 - The Rise of Digital Reading Past Hsc Exam Papers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Past Hsc Exam Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - $\circ\,$ Features to Look for in an Past Hsc Exam Papers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Past Hsc Exam Papers
 - Personalized Recommendations

- Past Hsc Exam Papers User Reviews and Ratings
- Past Hsc Exam Papers and Bestseller Lists
- 5. Accessing Past Hsc Exam Papers Free and Paid eBooks
 - Past Hsc Exam Papers Public Domain eBooks
 - Past Hsc Exam Papers eBook Subscription Services
 - Past Hsc Exam Papers Budget-Friendly Options
- 6. Navigating Past Hsc Exam Papers eBook Formats
 - o ePub, PDF, MOBI, and More
 - o Past Hsc Exam Papers Compatibility with Devices
 - Past Hsc Exam Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Past Hsc Exam Papers
 - Highlighting and Note-Taking Past Hsc Exam Papers
 - Interactive Elements Past Hsc Exam Papers
- 8. Staying Engaged with Past Hsc Exam Papers
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Past Hsc Exam Papers
- 9. Balancing eBooks and Physical Books Past Hsc Exam Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Past Hsc Exam Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Past Hsc Exam Papers
 - Setting Reading Goals Past Hsc Exam Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Past Hsc Exam Papers
 - Fact-Checking eBook Content of Past Hsc Exam Papers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web mark scheme gcse science a biology bl1fp june 2014 7 of 15 question answers extra information mark ao spec ref 2 a i in the direction of the force of gravity 1 ao2 1 2 3a 2 a ii against the force of gravity 1 ao2 1 2 3a 2 b i 1 2 3a diagram completed to show stem bending leaning towards the window

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web advice in all calculations show clearly how you work out your answer centre number candidate number surname other names candidate signature for examiner s use examiner s initials question mark 1 2 3 4 5 6 7 8 9 10 total f a 2 02 g jun14

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gcse science a biology mark scheme unit 01 biology january - Oct 19 2023

web mark scheme general certificate of education biology bl1fp january 2013 bl1fp question 4 question answers extra information mark 4 a i 2 two allow f and g 1 4 a ii c only 1 4 b any two from balanced diet otherwise malnourished release energy build cells growth repair ignore reference to health

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web teststreamsblog 2 min 31605 these biology questions and answers were pulled from our book biology questions for ss 1 compiled to serve as a reference material to help teachers draw up test and exam questions faster it could also help students assess their level of exam preparation each sample question includes correct answers

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web mark scheme gcse biology bl1fp june 2017 4 quality of written communication and levels marking in question 9 b students are required to produce extended written material in english and will be assessed on the quality of their written communication as well as the standard of the scientific response students will be

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web of scientific literacy the h1 biology syllabus is distilled from the h2 biology syllabus and key changes to the h1 science curriculum are in tandem with the changes in the h2 science curriculum the biology syllabus is developed as a seamless continuum from o level to a level without the need for topics to be revisited at a level

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web science a bl1fp unit biology b1 biology unit biology b1 friday 5 june 2015 1 30 pm to 2 30 pm for this paper you must have a ruler you may use a calculator time allowed 1 hour instructions use black ink or black ball point pen fill in the boxes at the top of this page answer all questions

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web science a biology bl1fp mark scheme 4405 4401 june 2016 version 1 0 final mark scheme mark schemes are prepared by the lead assessment writer and considered together with the total marks available for the question the typical answer or answers which are expected

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web 2 do not write outside the box 02 g k93070 jun13 bl1fp answer all questions in the spaces provided 1 the photograph shows an athlete at the start of a race 1 a the athlete s sense organs contain special cells these special cells detect changes in the environment 1 a i lists shows changes in the environment list b shows some of the

aqa bl1fp qp jun15 studocu - Sep 06 2022

web 4 b ii what type of cell is cell draw a ring around the correct answer 1 mark an egg cell a skin cell a sperm cell 4 b iii use the correct answer from the box to complete the sentence 1 mark cell membrane cytoplasm nucleus