| 3-4 | a) How 35Cl <sub>17</sub> and 35Cl <sub>17</sub> are related to each other?  | 5 |
|-----|--|---|
|     | b) If both of these forms of Cl occurs in nature as 75% and 25% respectively. Calculate the average atomic mass of Chlorine atom.  c) Draw the labelled schematic atomic structure of Chlorine atom having Z=17.   |   |
|     |  |   |
|     | Or   |   |
|     | <ul> <li>a) Distinguish between the properties of electrons, protons, and neutrons with respect to<br/>position, charge and mass.</li> </ul>   |   |
|     | b) Write the valency and atomic symbol of the following element:   |   |
|     | Sodium (Z=11)  |   |
|     | ii. Aluminium (Z=13)   |   |
|     | iii. Potassium (Z=19)  |   |
|     | iv. Calcium (Z=20)   |   |
| 3.5 | a) Plastids are present only in the plant. Which types of plastids are present in :  | 5 |
|     | i. Petals; ii. Leaves; iii. fruit pulp   |   |
|     | b) What happen if chloroplast is removed from the plant. Explain.  |   |
|     | Or   |   |
|     | To test the effect of the concentration of the solution on the cell a student placed some plant cell and animal cell (RBC) for 15 minute in the hypertonic solution and then shifted them into hypotonic solution. |   |
|     | Answer the following question:   |   |
|     | a) What is hypotonic solution.   |   |
|     | b) What will the expected observation in the above situation.  |   |
|     | c) what is the role of cell wall in the cell in above situation.   |   |
| 36  | a) Explain universal law of gravitation also write its mathematical expression.  | 5 |
|     | b) Write three importance of this law.   |   |
|     | Or   |   |
|     | Give reason:   |   |
|     | a) carry a bag with broader strap is easy than a thin strap bag.   |   |
|     | [  |   |
|     | b) why sharp knife cut easy in comparison with the blunt knife.  |   |

#### SECTION-E

Q.no.37 to 39 are case based / source- based questions with 2 to 3 short sub - parts.

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