

Review Sheet – Chemistry, Level 3 – Ch. 14: Liquids & Solids

Name: _____ ANSWERS _____

Date: _____

Period: _____

This test covers materials in the textbook in Chapters 14 (pp. 456–491). You are always responsible for the topics on the previous tests.

Test Expectations

The student should be able:

- *Describe how the Kinetic-Molecular Theory of Matter predicts the properties of substances*
- *Analyze the forces and relative strengths of the forces between particles in a solid, liquid or gas and compare the substances in terms of basic properties such as melting point, boiling point, viscosity, surface tension, or vapor pressure*
- *Describe the type and properties of one of the four solids*
- *Draw or use phase diagrams or heating curves to describe phase changes*

1. Define each of the following terms clearly and completely.

(a) condensed state of matter –

(b) phase –

(c) phase diagram –

(d) heating curve –

(e) intramolecular force –

(f) intermolecular force –

(g) metallic bond –

(h) hydrogen bond –

(i) dipole-dipole force –

(j) dispersion force –

(k) permanent dipole –

(l) temporary dipole –

(m) induced dipole –

(n) viscosity –

(o) surface tension –

(p) water –

(r) crystalline solid –

(s) amorphous solid

(t) adhesion –

(u) cohesion –

(v) covalent-network solid –

(w) phase change –

(x) vapor pressure –

(y) heat of fusion –

(z) heat of vaporization –

(aa) sublimation –

(bb) deposition –

(cc) vaporization –

(dd) condensation –

(ee) phase change –

(ff) freezing –

(gg) melting –

(hh) Kinetic-Molecular Theory of Matter –

Review Sheet Chemistry A Study Of Matter

Kerry L. Hull, Barbara Janson Cohen



Review Sheet Chemistry A Study Of Matter:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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Study Guide and Full Solutions Manual Susan McMurry,John McMurry,2002-08 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all end of chapter problems in the textbook

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