

CHAPTER 24: NUCLEONICS

Nuclear Chemistry

Module: Problems

Problem 1: Applications Answer the following questions in the space provided.

1. The radioactive isotope ^{238}Pu is used as a power source for spacecraft. The half-life of ^{238}Pu is 87.7 years. Calculate the mass of ^{238}Pu required to produce a power source that will last for 10 years. (Assume the power source is a constant power source.)

2. A typical nuclear reactor produces 10^6 W of power. Calculate the mass of ^{235}U required to produce this power. (Assume the reactor is 100% efficient.)

3. Write the nuclear equations for the following reactions:

a. ^{238}Pu decays to ^{234}U by emitting an α particle.

b. ^{235}U decays to ^{231}Th by emitting an α particle.

c. ^{235}U decays to ^{231}Th by emitting an α particle.

4. Write the complete nuclear equations for the following reactions:

a. ^{235}U decays to ^{231}Th by emitting an α particle.

b. ^{235}U decays to ^{231}Th by emitting an α particle.

c. ^{235}U decays to ^{231}Th by emitting an α particle.

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IM Harris



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