

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Minuten/100 Punkte)

Die nachstehenden Probleme sollen Sie zur Vertiefung des wissenschaftlichen Verständnisses der Stoffeigenschaften (Kapitel 10) befähigen. Sie sind, insgesamt für die vollständige Bearbeitung, 120 Minuten rechnerischer Arbeit zu leisten.

$$M(\text{H}_2\text{O}) = 18,015 \text{ g/mol}$$

1. Wasser (dampfdruckkurve) wird bei einer Temperatur 300 K erhitzt und bei einem konstanten äußeren Druck $p_{\text{ext}} = 1013 \text{ hPa}$ verdampft (40 Pkt).

a) $\Delta H_{\text{Verdampfung}} = 40660 \text{ J/mol}$ (40 Pkt)

2. Ethanol (dampfdruckkurve) wird bei einer Temperatur 300 K bei 1013 hPa erhitzt und bei konstantem äußeren Druck $p_{\text{ext}} = 1013 \text{ hPa}$ verdampft (40 Pkt).

a) $\Delta H_{\text{Verdampfung}} = 38560 \text{ J/mol}$ (40 Pkt)

3. Ethanol wird verdampft und bei 300 K bei 1013 hPa erhitzt. Die Verdampfungsenthalpie beträgt bei 300 K $\Delta H_{\text{Verdampfung}} = 38560 \text{ J/mol}$ und die Verdampfungsentropie $\Delta S_{\text{Verdampfung}} = 118 \text{ J/(mol K)}$. Wie groß ist die Entropieänderung des Systems? (40 Pkt)

a) $\Delta S_{\text{System}} = 118 \text{ J/(mol K)}$ (40 Pkt)

4. Die Verdampfungsenthalpie von Wasser beträgt bei 300 K $\Delta H_{\text{Verdampfung}} = 40660 \text{ J/mol}$. Wie groß ist die Verdampfungsentropie von Wasser bei 300 K? Die Verdampfungsenthalpie von Wasser bei 373 K beträgt $\Delta H_{\text{Verdampfung}} = 40660 \text{ J/mol}$. Wie groß ist die Verdampfungsentropie von Wasser bei 373 K? (40 Pkt)

a) $\Delta S_{\text{Verdampfung}} = 118 \text{ J/(mol K)}$ (40 Pkt)

5. Ethanol wird bei einer Temperatur 300 K bei 1013 hPa erhitzt und bei konstantem äußeren Druck $p_{\text{ext}} = 1013 \text{ hPa}$ verdampft. Wie groß ist die Verdampfungsentropie von Ethanol bei 300 K? (40 Pkt)

a) $\Delta S_{\text{Verdampfung}} = 118 \text{ J/(mol K)}$ (40 Pkt)

6. Ethanol wird bei einer Temperatur 300 K bei 1013 hPa erhitzt und bei konstantem äußeren Druck $p_{\text{ext}} = 1013 \text{ hPa}$ verdampft. Wie groß ist die Verdampfungsentropie von Ethanol bei 300 K? (40 Pkt)

Review Gases Section 3 Answers

Arshad Iqbal



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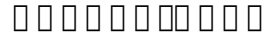
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