

PHYSICAL AND CHEMICAL CHANGES LAB- virtual

Pre-Lab Discussion:

Chemistry is the study of matter and the changes it undergoes. These changes can be divided into 2 classes: physical changes and chemical changes.

In a physical change, one or more physical properties of a substance are altered. Examples of such physical properties are: size, shape, color, and state of matter. Grinding, melting, dissolving, and evaporating are all physical changes. No new substance or substances are formed as a result of a physical change.

A chemical change results in the formation of one or more "new" substances. These new substances differ in chemical properties and composition from the original substance. Iron rusting and paper burning are two examples of chemical change. 5 signs of a chemical change are formation of bubbles (gas produced), formation of a precipitate (solid formed), change of color, change of temperature or production of an odor.

This experiment will help you understand the differences between physical and chemical changes and to recognize each type of change when it occurs.

Station 1 video

1. Scrape off a small amount of candle wax into a glass container.
2. Heat the outside of the glass container with a flame lighter until the wax melts. [video for wax scrape and heat](#)
3. Light the candle and allow it to burn for 30 seconds. [video of candle burning](#)

	observations	P or C	Evidence to support P or C
candle	its white its a solid	P	because it's still wax but just smaller pieces
Scraping of candle wax	melted fast, liquid, clear	P	it's still wax but there was a small color change and changed to a liquid
Melting of candle wax	it burned quickly, where did the wax go?, as it burned it became smaller	C	not reversible, its a "new" substance
Burning of candle	same this ^	C	not reversible, its a "new" substance

Station 2

1. Tear a piece of paper into pieces.
2. Place the pieces of paper on wire gauze and ignite them with a match. [video for paper tear and burn](#)

	observations	P or C	Evidence to support P or C
Sheet of paper	was white		
Torn paper	white, smaller	P	small pieces but still paper
Burned paper	ashes, color change,	C	turned to ash, cant g back, smoke

Prentice Hall Chemistry Physical And Chemical Change

Christian Drosten



Prentice Hall Chemistry Physical And Chemical Change:

Prentice Hall Chemistry, 2000 **Chemical Kinetics** Kenneth Antonio Connors, 1990 Chemical Kinetics The Study of Reaction Rates in Solution Kenneth A Connors This chemical kinetics book blends physical theory phenomenology and empiricism to provide a guide to the experimental practice and interpretation of reaction kinetics in solution It is suitable for courses in chemical kinetics at the graduate and advanced undergraduate levels This book will appeal to students in physical organic chemistry physical inorganic chemistry biophysical chemistry biochemistry pharmaceutical chemistry and water chemistry all fields concerned with the rates of chemical reactions in the solution phase **Introduction to Combustion**

Phenomena Kanury A Murty, 1975-12-15 *Atomic and Molecular Collision Theory* Franco A. Gianturco, 2012-12-06 Until recently the field of atomic and molecular collisions was left to a handful of practitioners who essentially explored it as a branch of atomic physics and gathered their experimental results mainly from spectroscopy measurements in bulk But in the past ten years or so all of this has dramatically changed and we are now witnessing the rapid growth of a large body of research that encompasses the simplest atoms as well as the largest molecules that looks at a wide variety of phenomena well outside purely spectroscopic observation and that finds applications in an unexpectedly broad range of physico chemical and physical processes The latter are in turn surprisingly close to very important sectors of applied research such as the modeling of molecular lasers the study of isotope separation techniques and the energy losses in confined plasmas to mention just a few of them As a consequence of this healthy state of affairs greatly diversified research pathways have developed however their specialized problems are increasingly at risk of being viewed in isolation although they are part of a major and extended branch of physics or chemistry This is particularly true when it comes to the theory of this work where well established methods and models of one subfield are practically unknown to researchers in other subfields and consequently the danger of wasteful duplication arising is quite real **Thermodynamics with Chemical Engineering Applications**

Elias I. Franses, 2014-08-25 Master the principles of thermodynamics and understand their practical real world applications with this deep and intuitive undergraduate textbook *Fundamentals of Metallurgical Processes* Lucien Coudurier, Donald W. Hopkins, Igor Wilkomirsky, 2013-10-22 Fundamentals of Metallurgical Processes Second Edition reviews developments in the design control and efficiency of metallurgical processes Topics covered include thermodynamic functions and solutions as well as experimental and bibliographical methods heterogeneous reactions metal extraction and iron and steelmaking This book is comprised of eight chapters and begins with an overview of the fundamentals of thermodynamics functions relationships and behavior of solutions followed by a discussion on methods of obtaining thermodynamic data from tables and graphs and by experiment The kinetics of heterogeneous reactions in metallurgy are examined next with particular reference to heterogeneous catalysis and mass transfer between immiscible liquid phases The following chapters focus on the extraction of metals from oxides sulfides and halides the production of iron and steel the structure and properties of slags

slag metal reactions and equilibria in iron and steel production The final chapter consists entirely of solved problems This monograph will be of interest to metallurgists and materials scientists **Using the Engineering Literature** Bonnie A. Osif, 2016-04-19 With the encroachment of the Internet into nearly all aspects of work and life it seems as though information is everywhere However there is information and then there is correct appropriate and timely information While we might love being able to turn to Wikipedia for encyclopedia like information or search Google for the thousands of links

Encyclopedia of Geomorphology Andrew Goudie, 2013-04-15 Geomorphology the discipline which analyzes the history and nature of the earth's surface deals with the landforms produced by erosion weathering deposition transport and tectonic processes In recent decades there have been major developments in the discipline and these are reflected in this major Encyclopedia the first such reference work in the field to be published for thirty five years Encyclopedia of Geomorphology has been produced in association with the International Association of Geomorphologists IAG and has a truly global perspective The entries have been written by an international editorial team of contributors drawn from over thirty countries who are all among the leading experts in the discipline In two lavishly illustrated volumes Encyclopedia contains nearly 700 alphabetically organized entries to provide a comprehensive guide both to specific landforms and to the major types of geomorphological processes that create them The Encyclopedia also demonstrates the major developments that have taken place in recent years in our knowledge of tectonic and climatic changes and in the use of new techniques such as modelling remote sensing and process measurement Older concepts however are not forgotten and provide an historical perspective on the development of ideas Both accessible and authoritative Encyclopedia of Geomorphology is destined to become the definitive resource for students researchers and applied practitioners in the field of geomorphology and the cognate disciplines of geography earth science sedimentology and environmental science **Hydrometallurgy** Michael L. Free, 2013-08-06 As the first book to compile the fundamentals applications reference information and analytical tools on the topic Hydrometallurgy presents a condensed collection of information that can be used to improve the efficiency and effectiveness with which metals are extracted recovered manufactured and utilized in aqueous media in technically viable and reliable environmentally responsible and economically feasible ways Suitable for students and researchers this college level overview addresses Fundamentals of Chemical Metallurgy in Aqueous Media Speciation and Phase Diagrams Rate Processes in Aqueous Metal Processing Aqueous Metal Extraction and Leaching Fundamentals of Metal Concentration Processes and more **Chemistry, 1996-2000** Ingmar Grenthe, 2003 A collection of the Nobel Lectures delivered by the prizewinners in chemistry together with their biographies portraits and the presentation speeches **Voet's Principles of Biochemistry** Donald Voet, Charlotte W. Pratt, Judith G. Voet, 2018 Voet's Principles of Biochemistry Global Edition addresses the enormous advances in biochemistry particularly in the areas of structural biology and bioinformatics It provides a solid biochemical foundation that is rooted in chemistry to prepare students for the scientific challenges of the future New

information related to advances in biochemistry and experimental approaches for studying complex systems are introduced. Notes on a variety of human diseases and pharmacological effectors have been expanded to reflect recent research findings. While continuing in its tradition of presenting complete and balanced coverage, this Global Edition includes new pedagogy and enhanced visuals that provide a clear pathway for student learning.

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Kinetics of Chemical Reactions Guy B. Marin, Gregory S. Yablonsky, 2011-08-29 This systematic presentation covers both experimental and theoretical kinetic methods as well as fundamental and applied. The identification of dominant reaction paths, reaction intermediates, and rate determining steps allows a quantification of the effects of reaction conditions and catalyst properties, providing guidelines for catalyst optimization. In addition, the form in which the equations are presented allows for their straightforward implementation for scale up and chemical reactor design purposes. Throughout, the methodologies given are illustrated by many examples.

Handbook of Infrared Optical Materials Paul Klocek, 2017-09-04 This book includes a comprehensive presentation of the fundamental physics of optical matter, the definition of material physical properties, the listing and comparison of the physical properties of infrared optical materials, and the theory, design, and survey of infrared optical coatings.

Investigating Chemistry Matthew Johll, 2006-03-17 Matthew Johll's book introduces students from a non-science background to the fundamentals of chemistry through an array of examples and applications from real life crime scenes, Sherlock Holmes stories, and authentic accounts of drug deals, murders, and thefts.

Combustion Theory Forman A. Williams, 2018-03-05 Combustion Theory delves deeper into the science of combustion than most other texts and gives insight into combustions from a molecular and a continuum point of view. The book presents derivations of the basic equations of combustion theory and contains appendices on the background of subjects of thermodynamics, chemical kinetics, fluid dynamics, and transport processes. Diffusion flames, reactions in flows with negligible transport, and the theory of pre-mixed flames are treated, as are detonation phenomena, the combustion of solid propellants, and ignition, extinction, and flammability phenomena.

The Cardiff Libraries Review, 1910

Fundamentals of Water Treatment Unit Processes David Hendricks, 2016-04-19 Carefully designed to balance coverage of theoretical and practical principles, *Fundamentals of Water Treatment Unit Processes* delineates the principles that support practice using the unit processes approach as the organizing concept. The author covers principles common to any kind of water treatment, for example, drinking water, municipal wastewater.

Modern Sample Preparation for Chromatography, 2014-10-18 Sample preparation is applied to make real world samples amenable for chromatographic analysis or to improve the results of this type of analysis. A wide variety of procedures are applied for this purpose, and their description is the main goal of the present book. The principles of these procedures are explained, discussing their advantages and disadvantages and their applicability to different types of samples, as well as their fit for different types of chromatographic analysis. This provides a guide for choosing the appropriate sample preparation for a given analysis. The book also contains numerous literature references and examples of sample preparation for different

matrices The material is presented in three parts one discussing physical methods used in sample preparation such as filtration distillation solvent extraction solid phase extraction electro separations Presents in a systematic way numerous techniques applied for sample preparation for chromatographic analysis Provides an up to date source of information regarding the progress made in sample preparation for chromatography Describes examples for specific type of matrices providing a guide for choosing the appropriate sample preparation method for a given analysis

Enzyme Kinetics: Catalysis and Control Daniel L. Purich, 2010-06-16 Far more than a comprehensive treatise on initial rate and fast reaction kinetics this one of a kind desk reference places enzyme science in the fuller context of the organic inorganic and physical chemical processes occurring within enzyme active sites Drawing on 2600 references Enzyme Kinetics Catalysis Control develops all the kinetic tools needed to define enzyme catalysis spanning the entire spectrum from the basics of chemical kinetics and practical advice on rate measurement to the very latest work on single molecule kinetics and mechanoenzyme force generation while also focusing on the persuasive power of kinetic isotope effects the design of high potency drugs and the behavior of regulatory enzymes Historical analysis of kinetic principles including advanced enzyme science Provides both theoretical and practical measurements tools Coverage of single molecular kinetics Examination of force generation mechanisms Discussion of organic and inorganic enzyme reactions

Thermodynamics and Kinetics for the Biological Sciences Gordon G. Hammes, 2000-06-26 Gain a working knowledge of thermodynamics and kinetics with a minimum of mathematics a guide for individuals in the biological sciences An understanding of thermodynamics and kinetics is essential for researchers investigating molecular phenomena in diverse disciplines including bioorganic chemistry medicinal chemistry biochemistry pharmaceuticals and biology The use of these physical chemistry tools in the biological sciences has exploded over the past fifteen years but the majority of works on thermodynamics and kinetics require mathematical expertise beyond that of many researchers in the field Presenting a highly accessible introduction to thermodynamics and kinetics Thermodynamics and Kinetics for the Biological Sciences employs a minimum of mathematics assuming only a basic calculus background while treating a wide range of topics in a logical and easy to follow style All principles and concepts are clearly illustrated through the use of relevant applications and examples from the biological sciences and explanations are further enhanced with problems and up to date references Written by a world renowned authority on biochemical kinetics this remarkable book also features an easy to understand statistical development of entropy and a more extensive coverage of chemical kinetics and ligand binding to macromolecules than is usually found in books of this kind Readers will acquire a working knowledge of thermodynamics and kinetics that they can readily apply to biological systems and use for exploring the scientific literature

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