

Chemistry - Review Chapters (415) 6

Directions: Answer each of the following questions. You need not use complete sentences.

1. By what property did Mendeleev arrange the elements?

ATOMIC MASS

2. By what property did Moseley suggest that the periodic table be arranged?

ATOMIC NUMBER

3. What is a period? How many are there in the periodic table?

A HORIZONTAL ROW IN THE PERIODIC TABLE; 7

4. What is a group (also called a family)? How many are there in the periodic table?

A VERTICAL COLUMN IN THE PERIODIC TABLE; 18

5. State the number of valence electrons in an atom of:

a. sulfur	b. calcium	c. chlorine	d. arsenic
6	2	7	5

6. Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10	b. 18	c. 36	d. 90
Ne, NEON	Ar, ARGON	Kr, KRYPTON	Th, THORIUM

7. List, by number, both the period and group of each of these elements.

	<u>Symbol</u>	<u>Period</u>	<u>Group</u>
a. beryllium	Be	2	2
b. iron	Fe	4	8
c. lead	Pb	6	14

8. Which of the following pairs of elements belong to the same period?

a. Na and Cl	b. Na and Li	c. Na and Cu	d. Na and Ne
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9. Which of the following pairs of elements belong to the same group?

a. H and He	b. Li and Be	c. C and Pb	d. Ga and Ge
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10. How does an element's period number relate to the number of the energy level of its valence electrons?

PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

11. What are the transition elements?

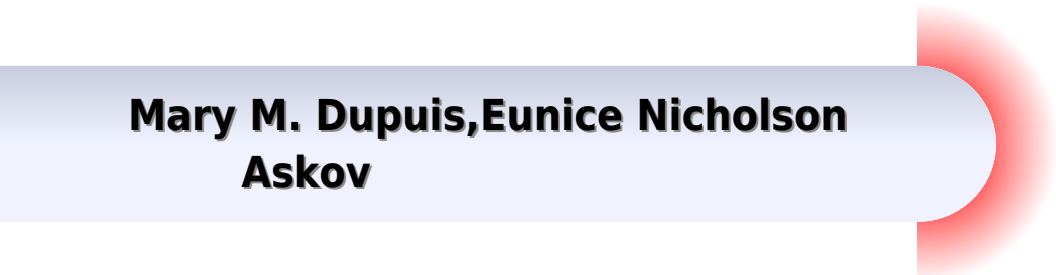
GROUPS 3-12

12. In what type of orbitals are the actinide and lanthanide electrons found?

f ORBITALS

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**Mary M. Dupuis, Eunice Nicholson
Askov**



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1. Understanding the eBook Prentice Hall Chemistry Chaopter 6 Test Answers
 - The Rise of Digital Reading Prentice Hall Chemistry Chaopter 6 Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry Chaopter 6 Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Chemistry Chaopter 6 Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry Chaopter 6 Test Answers
 - Personalized Recommendations
 - Prentice Hall Chemistry Chaopter 6 Test Answers User Reviews and Ratings
 - Prentice Hall Chemistry Chaopter 6 Test Answers and Bestseller Lists
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 - Prentice Hall Chemistry Chaopter 6 Test Answers Enhanced eBook Features
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 - Highlighting and Note-Taking Prentice Hall Chemistry Chaopter 6 Test Answers
 - Interactive Elements Prentice Hall Chemistry Chaopter 6 Test Answers
8. Staying Engaged with Prentice Hall Chemistry Chaopter 6 Test Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Chemistry Chaopter 6 Test Answers
9. Balancing eBooks and Physical Books Prentice Hall Chemistry Chaopter 6 Test Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Chemistry Chaopter 6 Test Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Prentice Hall Chemistry Chaopter 6 Test Answers
- Setting Reading Goals Prentice Hall Chemistry Chaopter 6 Test Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Prentice Hall Chemistry Chaopter 6 Test Answers
- Fact-Checking eBook Content of Prentice Hall Chemistry Chaopter 6 Test Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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