#### Chapter 2 The Chemistry of Life

## Section 2-1 The Nature of Matter (pages 35-39)

C Key Concepts

- What three subatomic particles make up atoms?
- How are all of the isotopes of an element similar?
- · What are the two main types of chemical bonds?

Atoms (page 35)

- The basic unit of matter is called a(an) \_\_\_\_\_\_
- Describe the nucleus of an atom. \_\_\_\_\_\_
- Complete the table about subatomic particles.

#### SUBATOMIC PARTICLES

Particle	Charge	Location in Atom	
	Positive	0	
	Neutral		
	Negative		

4. Why are atoms neutral despite having charged particles? \_\_\_\_\_

## Elements and Isotopes (page 36)

- 5. What is a chemical element?
- 7. Atoms of the same element that differ in the number of neutrons they contain are known as \_\_\_\_\_
- 8. How are isotopes identified? \_\_\_\_\_
- Why do all isotopes of an element have the same chemical properties?

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**LP Steffe** 

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