

Chapter 7 Chemical Reactions

Section 7.1 Describing Reactions

(pages 192–198)

This section discusses the use of chemical equations and how to balance them. It also demonstrates the use of calculations in chemistry.

Reading Strategy (page 192)

Monitoring Your Understanding Preview the Key Concepts, topic headings, vocabulary, and figures in this section. List two things you expect to learn. After reading, state what you learned about each item you listed. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

What I Expect to Learn	What I Learned
Answers may vary. Possible answers: How to balance chemical equations	Answers may vary. Possible answers: An unbalanced equation can be balanced by changing the coefficients.
How to convert from mass to moles	The mass of a substance can be converted to moles by using the molar mass as a conversion factor.

Chemical Equations (pages 192–193)

- Is the following sentence true or false? The new substances formed as a result of a chemical reaction are called products. true
- Circle the letter of each sentence that is a correct interpretation of the chemical equation $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$.
 - Carbon and oxygen react and form carbon monoxide.
 - ☒ Carbon and oxygen react and form carbon dioxide.
 - Carbon dioxide yields carbon and oxygen.
 - ☒ The reaction of carbon and oxygen yields carbon dioxide.
- Is the following sentence true or false? The law of conservation of mass states that mass is neither created nor destroyed in a chemical reaction. true
- Circle the letter of the correct answer. According to the equation $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$, how many carbon atoms react with 14 molecules of oxygen to form 14 molecules of carbon dioxide?
 - 1
 - 7
 - ☒ 14
 - 28
- In the reaction represented by the equation $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$, the mass of carbon dioxide produced equals the total mass of carbon and oxygen that reacted.

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Lei Shi



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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Principles and Applications of Aquatic Chemistry** François M. M. Morel, Janet G. Hering, 1993-03-10 Presents aquatic chemistry in a way that is truly useful to those with diverse backgrounds in the sciences Major improvements to this edition include a complete rewrite of the first three background chapters making them user friendly There is less emphasis on mathematics and concepts are illustrated with actual examples to facilitate understanding *Molecular Modeling and Prediction of Bioactivity* Klaus Gundertofte, Fleming Steen Jørgensen, 2012-12-06 Much of chemistry molecular biology and drug design are centered around the relationships between chemical structure and measured properties of compounds and polymers such as viscosity acidity solubility toxicity enzyme binding and membrane penetration For any set of compounds these relationships are by necessity complicated particularly when the properties are of biological nature To investigate and utilize such complicated relationships henceforth abbreviated SAR for structure activity relationships and QSAR for quantitative SAR we need a description of the variation in chemical structure of relevant compounds and biological targets good measures of the biological properties and of course an ability to synthesize compounds of interest In addition we need reasonable ways to construct and express the relationships i e mathematical or other models as well as ways to select the compounds to be investigated so that the resulting QSAR indeed is informative and useful for the stated purposes In the present context these purposes typically are the conceptual understanding of the SAR and the ability to propose new compounds with improved property profiles Here we discuss the two latter parts of the SAR/QSAR problem i e reasonable ways to model the relationships and how to select compounds to make the models as good as possible The second is often called the problem of statistical experimental design which in the present context we call statistical molecular design SMD 1 *Physical Ceramics for Engineers* Lawrence H. Van Vlack, 1964
Prentice Hall Earth Science, 1988 *Focus on Earth Science*, 2001 *Chemistry for the Life Sciences* J. G. Dawber, Alan Thomas Moore, 1973 Chemistry For the Life Sciences covers the basic chemistry of HNC HND courses in applied biology and medical laboratory subjects but the treatment is also appropriate to post HNC courses in biochemistry such as the M I Biol and to certain degree level courses The text is designed to provide a unified approach to the subject with roughly equivalent amounts of physical and organic chemistry and a small amount of inorganic chemistry in chapter one It proceeds from a relatively pure chemistry in the early chapters to aspects of biochemistry and molecular biology in later chapters Publisher *Polymer Science and Technology* Robert O. Ebewele, 2000-03-23 By consolidating into one volume the

fundamentals currently covered piecemeal across several reference this book simplifies the learning of polymer science Its primary focus is the ultimate property of the finished polymer product Part I explains polymer fundamentals Part II discusses how polymers are prepared from monomers and the transformation of polymers into useful everyday articles Part III examines the properties and applications of polymers Polymer Science and Technology presents these aspects of the science in a readily understandable way It emphasizes basic qualitative comprehension of concepts rather than their rote memorization or detailed mathematical analysis

Biology Leslie MacKenzie, David K. Arwine, Edward J. Shewan, Michael J. McHugh, 2004-08 Originally developed by the Creation Research Society this classic text is now available in an updated and full color edition This hardbound text contains helpful questions and a thorough presentation of biology concepts Beautiful graphs and illustrations complement the text material that is scientifically accurate and true to six day young earth creationism Grades 9-10

Air Quality, Fourth Edition Thad Godish, Joshua S. Fu, 2003-07-28 Ozone destroying chemicals greenhouse gases and dangerous airborne substances that were once thought to be benign are the most urgent issues facing air pollution control experts Students need a thorough updated reference that explores these current trends while also covering the fundamental concepts of this emerging discipline A new revision of a bestseller Air Quality Fourth Edition provides a comprehensive overview air quality issues including a better understanding of atmospheric chemistry the effects of pollution on public health and the environment and the technology and regulatory practices used to achieve air quality goals New sections cover toxicological principles and risk assessment The book also contains revised discussions on public policy concerns with a focus on air quality standards for ozone depletion and global warming and the health effects of particulate air pollutants This edition continues to serve as a very readable text for advanced level undergraduate and early graduate study in environmental science environmental management and in programs related to the study of public health industrial hygiene and pollution control

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including in situ studies The work is underpinned through an unprecedented structural perspective of molecular features solid state packing arrangements and surface energetics as well as in situ studies This work will be of interest to researchers industrialists intellectual property specialists and policy makers interested in the latest developments in the design and supply of advanced high added value organic solid form materials and product composites *Chemistry* Henry Dorin,1987

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