

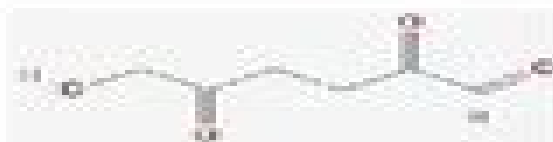
A) square planar B) bent see-saw C) trigonal pyramidal D) trigonal planar E)

Short Answer:



1a. How many pi bonds and sigma bonds are there in the molecule above? (2T)

# of sigma bonds in the molecule	
# of pi bonds in the molecule	



2b. What are the bond angles and hybridization of the oxygen of the O-H group in the molecule above? (2T)

question	angles	hybridization
2b.		

2c. What are the bond angles and hybridization of the carbon of the aldehyde in the molecule above? (2T)

question	angles	hybridization
2c.		

Prentice Hall Chemistry Covalent Bonding Test Answers

SB Merriam

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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In a digitally-driven earth where monitors reign supreme and immediate connection drowns out the subtleties of language, the profound techniques and emotional subtleties concealed within phrases usually go unheard. Yet, situated within the pages of **Prentice Hall Chemistry Covalent Bonding Test Answers** a fascinating fictional value pulsing with fresh feelings, lies an exceptional journey waiting to be undertaken. Written by an experienced wordsmith, this charming opus invites visitors on an introspective trip, softly unraveling the veiled truths and profound affect resonating within the very fabric of each word. Within the mental depths with this touching review, we can embark upon a heartfelt exploration of the book is key themes, dissect their charming writing model, and fail to the strong resonance it evokes serious within the recesses of readers hearts.

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