

Periodic and Groups: Valence Electrons: Mastery

Periodic and Groups

Elements that have the same number of valence electrons have similar properties, so we put them in groups. The the periodic table groups + columns that look like this. They will see family groups 1A-18A. Hydrogen is in the same period as Helium.

Elements that have electrons in the outer shell have similar properties in the Periodic Table. Periods go left to right. Elements in the same vertical columns

Period	Group		Group					18A	
	1A	2A	1A	2A	3A	4A	5A	6A	7A
1	H 1H	He 2He							
2	Li 3Li	Be 4Be							
3	Na 11Na	Mg 12Mg	Transition metals in between these rows						
4	K 19K	Ca 20Ca							
5	Rb 37Rb	Sr 38Sr							
6	Cs 55Cs	Ba 56Ba							
7	Fr 87Fr	Ra 88Ra							

Thinking like the game "Monopoly", answer that can describe the period and group.

What element is in Group 1, and Period 10? (Group 1 is not in 10th period. It is in 7th period.)

What element is in Group 10, and Period 1?

What element is in Group 1A, and Period 10?

What group number is in Period 10?

What group number is in Period 10?

Valence Electrons

Valence Electrons are the most important electrons in an atom. Each period contains the same number of valence electrons. Valence electrons are the electrons that are involved in chemical bonding.

Hydrogen 1



1 valence electron

Helium 2



2 valence electrons

Lithium 3



3 valence electrons

Hydrogen and Helium are both in Group 1A. Helium is 2 valence electrons. Hydrogen is 1 valence electron.

Group 1A, Helium 2He and Group 1A, Lithium 3Li. How many valence electrons do they have?

How many valence electrons?

Group 1A: Helium has 2 valence electrons. Lithium has 3 valence electrons. Group 2A: Helium has 2 valence electrons. Group 3A: Helium has 3 valence electrons. Group 4A: Helium has 4 valence electrons. Group 5A: Helium has 5 valence electrons. Group 6A: Helium has 6 valence electrons. Group 7A: Helium has 7 valence electrons. Group 8A: Helium has 8 valence electrons.

Periods And Groups Valence Electrons Masses Answers

Michael Seilmaier



Periods And Groups Valence Electrons Masses Answers:

Reviewing **Periods And Groups Valence Electrons Masses Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Periods And Groups Valence Electrons Masses Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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