

5 - Periodicity Activity

20 19 18 17 16 15 14 13 12 11 10 9 8 7 6 5 4 3 2 1

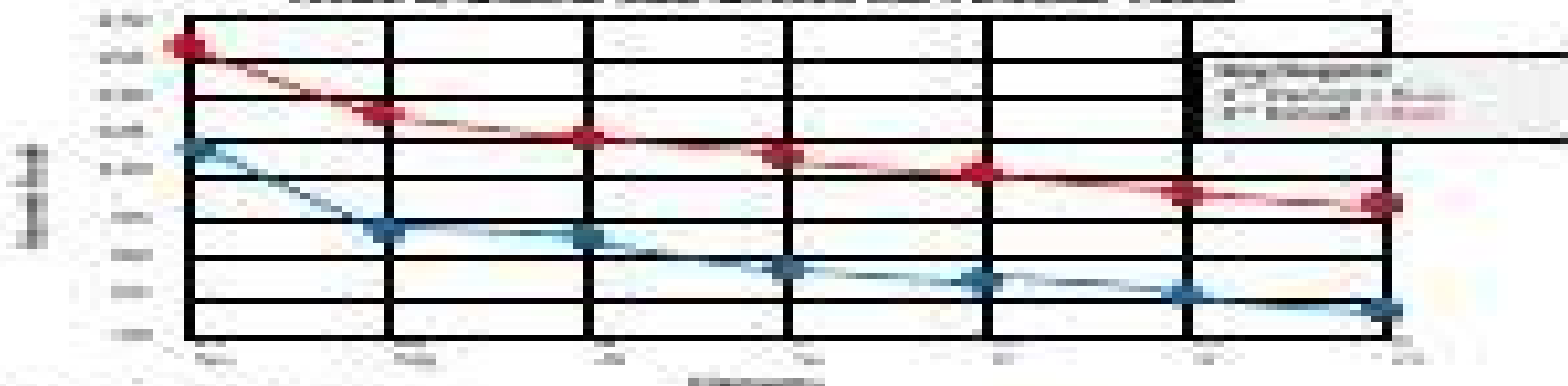
Instructions: Using the graphs above fill the boxes of atomic properties below:

1. Choose which graph best fits the property stated in the graphing sheet of 5 data points (graph above)
2. Is a slight downward trend in this graph Period? If not, use the other graphs
3. Label the box as 'not periodic' (highlighted)
4. Answer all questions...

	1	2	3	4	5	6	7	8
1	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius
2	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius
3	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius
4	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius
5	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius
6	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius
7	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius
8	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius	Electronegativity	Atomic Radius

Now label the trends in the following table:

Trend in Atomic Size across the Periodic Table



Electronegativity Questions

1. Following the trend in atomic size across the periodic table, Electronegativity
2. This element has been shown and is Electronegativity (highlighted) property. It is a non-metal (highlighted) element. It is a metalloid (highlighted) element.

Atomic Radius Questions

1. Electronegativity trend in the periodic table is downward (highlighted) i.e. from top-left (highlighted) to bottom-right (highlighted). It has an increasing (highlighted) trend in the periodic table.
2. This element has been shown and is Atomic Radius (highlighted) property. It is a metal (highlighted) element. It is a metalloid (highlighted) element.

Periodicity Activity Okstate Answer Key

Scott C. Dulebohn



Periodicity Activity Okstate Answer Key:

Reviewing **Periodicity Activity Okstate Answer Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Periodicity Activity Okstate Answer Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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