

Average Atomic Mass

How can the masses of the particles within an atom be determined?

Objectives

After observing three gases, three ions, and naturally occurring three isotopes, the masses of these particles, ions, and atoms should be determined. Comparison of particles including those belonging to the same element should show three different values resulting from isotopes of particles and elements having three different particles. Finally, the atomic mass of an element should be determined using the masses of the particles within the atom and the relative abundance of each particle within the atom.

Model 1 – A Strip of Magnesium Metal



1. What are the masses of the particles within the metal Mg shown in Model 1?
They are the same, they are all 24.
2. What are the masses of the particles within the metal Mg shown in Model 1?
24, 24, 24.
3. How all of the masses of the particles in Model 1? They are the same, they are all 24.
Yes, they are the same, they are all 24.
4. How the masses of the particles in Model 1? They are the same, they are all 24.
Yes, they are the same, they are all 24.

	^{24}Mg	^{25}Mg	^{26}Mg
Abundance (%)	75	25	0

5. What is the mass of the particles in the metal Mg shown in Model 1?
24.
6. What are the masses of the particles in the metal Mg shown in Model 1? They are the same, they are all 24.
Yes, they are the same, they are all 24.

Periodicity Of Elements Pogil Key

R Pring



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