#### Chemistry Periodic Table Test Chapter 6 Chem IH

Matching Match each item with the correct statement below.

		electronegativity     ionization energy     electrons     metal	e. f. g. h.	transition metal periodic law group period	
	1.0	vertical column in the periodic table			
	2				
_	3	그를 부르고 하면 가는 그렇게 하는 것이 되었다. 그런 그렇게 하는 것이 되었다. 그렇게 하는 것이 없는 것이 되었다. 그렇게 하는 것이 없는 것이었다면 없는 없는 것이었다면 없는 없는 것이었다면 없어요. 되었다면 없는 것이었다면 없는 것이었다면 없는 것이었다면 없는 것이었다면 없어요. 없어요 없는 것이었다면 없는 것이었다면 없다면 없어요. 없어요 없어요 없어요. 없어요 없어요 없어요. 없어요 없어요 없어요. 없어요 없어요 없어요. 없어요 없어요. 없어요 없어요 없어요. 없어요 없어요 없어요. 없어요 없어요. 없어요 없어요 없어요. 없어요 없어요 없어요. 없어요 없어요. 없어요 없어요 없어요. 없어요.			
		ability of an atom to attract electrons when the atom is in a compound			
-		The state of the s			
		energy required to remove an electron f	Potin am aron		
No.	ttiple C	Thoice			
		e letter of the choice that best completes t	he statemen	st or answers the question.	
	60	What is another name for the main grou	in elements	9	
		a. s block elements		s & p block elements	
		b. p block elements elements		d block elements	
	7.	What is another name for the transition	metals?		
		a. noble gases		p block elements elements	
		b. s block elements elements		d block elements	
	8.	Who arranged the elements according to missing elements?	o atomic ma	ass and used the arrangement to predict the properties of	
		a. Henry Moseley	·C.	John Dalton	
		b. Antoine Lavoisier	d.	Dmitri Mendeleev	
	90	Which of the following categories inclu	ides the maj	ority of the elements?	
		a. metalloids	c.	metals	
		b. liquids	d.	nonmetals	
	_ 10.	In which of the following sets is the syr electrons given correctly?	nbol of the	element, the number of protons, and the number of	
		a. In, 49 protons, 49 electrons	C.	Cs, 55 protons, 132.9 electrons	
		<ul> <li>b. Zn, 30 protons, 60 electrons</li> </ul>	di	F, 19 protons, 19 electrons	
	115	The atomic number of an element is the	total mamb	er of which particles in the nucleus?	
		a. neutrons	C.	electrons	
		b. protons		protons and electrons	
	1.2.	What element has the electron configur	ation 1s22s	$^{2}2p^{6}3s^{2}3p^{2}7$	
		a. nitrogen	e.	silicon	
		b. selenium	d.	silver	
	13.	Which subatomic particle plays the gre-	atest part in	determining the properties of an element?	
		a. proton		neutron	
		b. electron	d.	none of the above	
	1.4	Which of the following elements is a tra			
		a. cesium	C.		
		b. copper		tin	

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**M Lipman** 

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