

Percent Yield Homework

I performed the following chemical reaction:



- 1) Balance the equation in the spaces above.
- 2) What type of reaction is taking place here? _____
- 3) Complete the equation by showing symbols of state as well as symbols around the arrow and Δ values, if possible.
- 4) If I were to start with 40 grams of $\text{C}_2\text{H}_5\text{OH}$ and an excess of oxygen, how many grams of carbon dioxide might I expect to make?

- 5) If my actual yield of carbon dioxide was 25 grams, what is my percent yield for this reaction?

- 6) Is my answer from problem 4 reasonable? If it is, explain why. If not, explain why not.

- 7) Explain why chemical reactions very rarely have a percent yield of 100%.

Percent Yield Sheet

Ian Pickup



Percent Yield Sheet:

Percent Yield Sheet Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Percent Yield Sheet**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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