

## Chapter 1: Limiting Reagents and Percent Yield (Pages 1-4)

**Remember:** Limiting reagent is the reactant that will completely react and leave another reactant leftover. **Calculate:** All limiting reagent must limit everything that comes along. **Remember:** Limiting reagent always comes up first when figuring out.

Chemists will determine the limiting reagent by determining the amount of product that each reagent will form. The limiting reagent is the one that forms the least amount of product.



The reactant required with the least amount is the limiting reagent.

2H <sub>2</sub> (g)	100.00 grams
O <sub>2</sub> (g)	50.00 grams
H <sub>2</sub> O(l)	100.00 grams
H <sub>2</sub> (g)	100.00 grams

1. 100.00 g of each reactant will be used. 100.00 g of each reactant means 100.00 g of each reactant. **Remember:** Limiting reagent will be the one that will be used first.



2. **Remember:** Limiting reagent is the one that will be used first. **Remember:** Limiting reagent is the one that will be used first.

3. **Remember:** Limiting reagent is the one that will be used first. **Remember:** Limiting reagent is the one that will be used first.



4. **Remember:** Limiting reagent is the one that will be used first. **Remember:** Limiting reagent is the one that will be used first. **Remember:** Limiting reagent is the one that will be used first.



# Percent Yield And Limiting Reactants Quiz

**Raffaela Di Napoli**



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