

### Question 1: Russia autocracy and revolt

(a) Study these events which occurred in the years 1903–1912.

The split in the Social Democrat Party	The Lena Goldfield Strike	The assassination of Stolypin	The setting up of the Octobrists	The first duma
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Write these events in the correct chronological sequence.

(3)

- 1- The split in the Social Democrat Party
- 2- The setting up of the Octobrists
- 3- The first Duma
- 4- The assassination of Stolypin
- 5- Lena Goldfield strike

(b) Choose **either**

The Lena Goldfield strike

The first Duma

Explain **one** effect on Russia of the event you have chosen.

(4)

The Lena Goldfield strike refers to an incident where in 1912, the company food canteen attempted to distribute rotten horse meat disguised as beef. This, on top of other long-standing grievances such as long working hours and poor pay, sparked a spontaneous but widespread strike amongst the miners in which the government sent soldiers to deal with the uprising, resulting in the deaths of hundreds of workers. The effect of the strike was to re-ignite the ill-feelings of the ordinary Russian workers towards the Tsar. They felt that the Tsar was not willing to listen to the problems affecting them and used the same violent tactics as had been the case during Bloody Sunday in 1905. The result was that the Tsar lost even more support from ordinary workers.

(c) Why did Alexander III carry out a policy of repression in the years 1881–94?

(8)

One reason that Tsar Alexander III carried out a policy of repression was the influence of his key advisor Konstantin Pobedonostsev. He had been Alexander III's tutor and became a very powerful Prime Minister when the Tsar took power. Pobedonostsev's ideas were very conservative and he favoured strong autocratic rule as the best way to govern Russia. He was responsible for influencing Alexander III to assert himself at the start of his rule by crafting the Manifesto of Unshakable Autocracy in 1881. Pobedonostsev was therefore a key reason for Alexander III's policy of repression.

Pobedonostsev's influence to restore autocracy was further enhanced by the fact that the previous Tsar, Alexander II had been assassinated by a radical opposition terrorist group. The need to restore order and punish his father's killers led to Alexander III agreeing to harsh measures against the political opposition. This included the increased use of the secret police (the Okhrana) to arrest, imprison and torture anyone who threatened the Tsar's hold on power. The policy of repression was therefore partly caused by Tsar Alexander III's reaction to the assassination of his father.

The need to restore the Tsar's authority after the killing of the previous monarch also led to the imposition of strict measures to control the ethnic minorities which were also a threat to the Romanov regime. This was enforced through the policy of Russification which imposed Russian language and cultures on all people including non-Russians. The policy also included suppression of Jews through pogroms legalised by the May Laws. The need to reassert the government's control over ethnic minorities therefore played a role in the Tsar's policy of repression.

In conclusion, the most important reason for the introduction of the policy of repression was the manner of the death of Tsar Alexander II. This had a massive impact on Tsar Alexander III who had

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**Peipei Pang**



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web 33mol o<sub>2</sub> 6mol br<sub>2</sub> 11mol o<sub>2</sub> 18mol br<sub>2</sub> identify the limiting reactant s and excess reactant s the limiting reactant is o<sub>2</sub> since it would yield the least amount of product 18 mol br<sub>2</sub> the excess reactant is c<sub>2</sub>h<sub>3</sub>br<sub>3</sub> since its complete reaction would have yielded up to 54 mol br<sub>2</sub> think about your result

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o<sub>2</sub> 0.2398 g mgo vs 25.2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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web please save your changes before editing any questions 5 minutes 1 pt viewing model 3 if 4 moles of hydrogen react with 2 moles of oxygen which of these is the excess reactant is leftover at the end and how many moles of it are leftover hydrogen 2 moles leftover hydrogen 1 mole leftover

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web if you start with 14.82 g of ca(oh)<sub>2</sub> c a(oh)<sub>2</sub> and 16.35 g of h<sub>2</sub>so<sub>4</sub> h<sub>2</sub>so<sub>4</sub> a determine the limiting reagent b determine the number of moles of h<sub>2</sub>o h<sub>2</sub>o produced c determine the number of grams of caso<sub>4</sub> c a(oh)<sub>2</sub> produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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