

# REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*  
 S *sulphur*  
 Cl *chlorine*  
 C *carbon*  
 N *nitrogen*  
 Na *sodium*  
 Pb *lead*  
 Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*  
 phosphorus *P*  
 fluorine *F*  
 magnesium *Mg*  
 potassium *K*  
 calcium *Ca*  
 zinc *Zn*  
 iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79**      copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5**      calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

| Element      | Atomic Number | Atomic Mass | Protons   | Neutrons  | Electrons |
|--------------|---------------|-------------|-----------|-----------|-----------|
| <b>Lead</b>  | 82            | <b>207</b>  | <b>82</b> | 125       | <b>82</b> |
| <b>Argon</b> | <b>18</b>     | <b>38</b>   | 18        | 40        | <b>18</b> |
| Barium       | <b>56</b>     | 137         | <b>56</b> | <b>81</b> | <b>56</b> |
| <b>Tin</b>   | <b>50</b>     | 112         | <b>50</b> | <b>62</b> | 50        |

6. Match the unit with its description:

**F** 1. amount of substance  
**C** 2. density  
**B** 3. mass  
**D** 4. molar mass  
**A** 5. energy  
**E** 6. volume

A. joule  
 B. g  
 C. g/mL  
 D. g/mole  
 E. L  
 F. mole

# Review Answers Chemistry

**Michael Reid**



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