7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative positive crisscross noble gas electrons anion

octet ions cation electron configuration polyatomic empirical formula

- An ionic bond is an attraction between oppositely charged —ions
- Anions have a -negative ----- charge.
- An atom becomes an ion by losing or gaining ____electrons
- The _octet __rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
- When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) _noble gas
- 6. The atoms that make up **polyatomic** _ions are bonded together by covalent bonds.
- The _crisscross __method can be used to write the formula for an ionic compound.
- 8. The _empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

- losing . 9. Calcium becomes a monatomic cation by gaining two electrons.
- _ true _____ 10. A cation has a positive charge.
- _polyatomic __ 12. Monatomic ions consist of more than one atom.
- __seven _____13. The Lewis dot diagram for chlorine, a group 7A element, has aix electrons.
- __true (released) _ 14. A great deal of energy is produced when an ionic compound is formed.

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Table of Contents Review And Reinforcement Answers Prentice Hall

- 1. Understanding the eBook Review And Reinforcement Answers Prentice Hall
 - The Rise of Digital Reading Review And Reinforcement Answers Prentice Hall
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Review And Reinforcement Answers Prentice Hall
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - o Popular eBook Platforms
 - Features to Look for in an Review And Reinforcement Answers Prentice Hall
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Review And Reinforcement Answers Prentice Hall
 - Personalized Recommendations
 - Review And Reinforcement Answers Prentice Hall User Reviews and Ratings
 - Review And Reinforcement Answers Prentice Hall and Bestseller Lists
- 5. Accessing Review And Reinforcement Answers Prentice Hall Free and Paid eBooks
 - Review And Reinforcement Answers Prentice Hall Public Domain eBooks
 - Review And Reinforcement Answers Prentice Hall eBook Subscription Services
 - Review And Reinforcement Answers Prentice Hall Budget-Friendly Options
- 6. Navigating Review And Reinforcement Answers Prentice Hall eBook Formats

- o ePub, PDF, MOBI, and More
- Review And Reinforcement Answers Prentice Hall Compatibility with Devices
- Review And Reinforcement Answers Prentice Hall Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Review And Reinforcement Answers Prentice Hall
 - Highlighting and Note-Taking Review And Reinforcement Answers Prentice Hall
 - Interactive Elements Review And Reinforcement Answers Prentice Hall
- 8. Staying Engaged with Review And Reinforcement Answers Prentice Hall
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Review And Reinforcement Answers Prentice Hall
- 9. Balancing eBooks and Physical Books Review And Reinforcement Answers Prentice Hall
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Review And Reinforcement Answers Prentice Hall
- 10. Overcoming Reading Challenges
 - $\circ\,$ Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Review And Reinforcement Answers Prentice Hall
 - Setting Reading Goals Review And Reinforcement Answers Prentice Hall
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Review And Reinforcement Answers Prentice Hall
 - Fact-Checking eBook Content of Review And Reinforcement Answers Prentice Hall
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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