

Question 1: Section 11.1: Surface and Volume

11.1. Describing Surface-Volume and Properties

- ☐ a. physical
- ☐ b. chemical
- ☐ c. physical
- ☐ d. physical
- ☐ e. chemical
- ☐ f. physical

17. A substance whose volume expands or contracts appreciably with temperature has a proportional relationship of the volume to temperature.

- ☐ a. A substance whose volume keeps the relationship. The partial volume is not constant in a solid state.

- ☐ b. 1
- ☐ c. 2
- ☐ d. 3
- ☐ e. 4
- ☐ f. 5
- ☐ g. 6
- ☐ h. 7
- ☐ i. 8
- ☐ j. 9

11.2. Describing Matter States and Properties

- ☐ a. volume (volume) is not a state of matter. Answer
- ☐ b. density (mass/volume) is a state of matter. Answer
- ☐ c. pressure of the substance is the state of the matter. Answer

Based on the physical property, weight (mass) of a substance, answer the following questions. What is the state of the matter in which the state of matter is a solid state?

- ☐ a. A substance is a solid state when the volume of matter is 1.
- ☐ b. A substance is a solid state when the volume of matter is 2.
- ☐ c. A substance is a solid state when the volume of matter is 3.
- ☐ d. A substance is a solid state when the volume of matter is 4.
- ☐ e. A substance is a solid state when the volume of matter is 5.
- ☐ f. A substance is a solid state when the volume of matter is 6.
- ☐ g. A substance is a solid state when the volume of matter is 7.
- ☐ h. A substance is a solid state when the volume of matter is 8.
- ☐ i. A substance is a solid state when the volume of matter is 9.
- ☐ j. A substance is a solid state when the volume of matter is 10.

Based on the physical property, weight (mass) of a substance, answer the following questions. What is the state of the matter in which the state of matter is a solid state?

11.3. Describing Matter States and Properties

- ☐ a. physical and chemical

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**Dr. Gopal Jee Gopal, Ms. Prachi
Patel, Mr. Rudrank Moharana**



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