

16

SOLUTIONS

SECTION 16.1 PROPERTIES OF SOLUTIONS (pages 471–477)

This section describes the factors that affect the solubility of a substance and determines the rate at which it occurs dissolves.

► Solution Formation (pages 471–472)

Look at Figure 16.1 on page 471 to help you answer Questions 1 and 2.

- Underline the conditions that cause sugar to dissolve faster in water.
 - as a whole cube or as granulated sugar
 - when allowed to melt or when not melted
 - at a higher temperature or at lower temperature
- Name three factors that influence the rate at which a solute dissolves in a solvent.
 - _____
 - _____
 - _____
- In the following sentence, place an **all** or **few** particles dissolve more rapidly than larger particles because their particles expose a greater surface area to the colliding solvent molecules. _____

► Solubility (pages 472–473)

- Complete the following table, showing the steps in a procedure to determine the total amount of sodium chloride that will dissolve in 100 g of water at 25°C.

Procedure	Amount Dissolved	Amount Not Dissolved
Add 35.0 g of sodium chloride to the water.	35.0 g	0.0 g
Add an additional 1.0 g of sodium chloride.		
Determine the total amount that dissolved.		

- The amount of a substance that dissolves in a given quantity of solvent at a constant temperature is called the substance's _____ at that temperature.
- If a solution contains the maximum amount of solute for a given quantity of solvent at a constant temperature, it is called a(n) _____ solution.

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Mathematical Methods for Physical and Analytical Chemistry David Z. Goodson, 2011-11-14 Mathematical Methods for Physical and Analytical Chemistry presents mathematical and statistical methods to students of chemistry at the intermediate post calculus level The content includes a review of general calculus a review of numerical techniques often omitted from calculus courses such as cubic splines and Newton's method a detailed treatment of statistical methods for experimental data analysis complex numbers extrapolation linear algebra and differential equations With numerous example problems and helpful anecdotes this text gives chemistry students the mathematical knowledge they need to understand the analytical and physical chemistry professional literature *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright Office, 1959 Includes Part 1 Number 1 Books and Pamphlets Including Serials and Contributions to Periodicals January June **Core Media Collection for Secondary Schools** Lucy Gregor Brown, Betty McDavid, 1979 **Books and Pamphlets, Including Serials and Contributions to Periodicals** Library of Congress. Copyright Office, 1949 ,

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