



# Samsung Galaxy Mini S5570 Manual

**A Loxley**



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fusion instead of chemical equations where it shows the different number of elements is conserved in a reaction in a nuclear reaction the atomic mass and proton number are conserved

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web a nucleus changes into a new element by emitting nuclear radiations these changes are described using nuclear equations example  ${}^{86}_{219}\text{textrm{rn}} \rightarrow {}^{84}_{215}\text{textrm{po}}$

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web in alpha decay an alpha particle is ejected from an unstable nucleus so here s our unstable nucleus uranium 238 an alpha particle has the same composition as a helium nucleus we saw the helium nucleus in the previous video there are two protons in the helium nucleus and two neutrons

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web sep 12 2022 the half life of the nuclei and thus the material is  $t_{1/2} = 693 \text{ t}$   $t_{1/2} = 693 \text{ million years}$  35 a the decay constant is  $\lambda = 1/99105 \text{ s}^{-1}$   $\lambda = 1/99105 \text{ s}^{-1}$  b since strontium 91 has an atomic mass of 90 90 g

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