

Chemical Reactions Quiz (Quiz ID HMLJW2I)

1. The law that states that atoms do not appear or disappear in chemical reactions, is the Law of _____.

- ☐ (A) Conservation of Energy
- ☐ (B) Atom Stability
- ☐ (C) Conservation of Matter
- ☐ (D) Balancing Equations

2. In a chemical reaction you begin with 18 atoms. When the reaction is over you would expect to have _____ atoms remaining.

- ☐ (A) less than 18
- ☐ (B) 18
- ☐ (C) more than 18
- ☐ (D) Not enough info to determine this

3. Which statement correctly supports the argument that the illustrated chemical equation shown violates the law of conservation of matter?



- ☐ (A) There are more oxygen atoms on the left side of the arrow than on the right side
- ☐ (B) There are more hydrogen atoms on the left side of the arrow than on the right side.
- ☐ (C) There are more oxygen atoms on the right side of the arrow than on the left side.
- ☐ (D) There are more hydrogen atoms on the right side of the arrow than on the left side

4. Which is the most accurate description of the changes that take place in atoms when they form chemical bonds?

- ☐ (A) Atoms gain, share, or lose protons
- ☐ (B) Atoms gain, share, or lose neutrons
- ☐ (C) Atoms gain, share, or lose electrons
- ☐ (D) Atoms gain, share, or lose positrons

Quiz Section Describing Chemical Reactions

Robert R. Redfield



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