

LAB: CALCULATING A LIMITING REACTANT

Name _____

Lab Determination of a Limiting Reactant

Objective: Observe chemical reaction between magnesium metal and hydrochloric acid. Calculate the limiting and excess reactant.

Materials: 2 M HCl test tube

Procedure:

1. Record the mass of magnesium: _____ g
2. Place the test tube in a test tube rack. Add the magnesium strip.
3. Add 4 mL of 2 M HCl into the test tube.
4. Record your observations. How do you know a chemical reaction is taking place? Be sure to feel the side of the test tube.
5. Pour the contents into the waste beaker provided by your teacher. Rinse the rack upside down. You never want to pour unreacted solid.
6. Write the balanced equation for the reaction that you observed and balance the equation.



Quick Lab Limiting Reagents

SJ Ball



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Teaching Science for Understanding James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities

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Practical Lab Manual of Pharmaceutical Organic Chemistry - II Dr. Priyanka Gupta Manglik, 2024-08-15 Builds upon Volume I covering advanced organic chemistry experiments with relevance to drug synthesis mechanisms and qualitative quantitative analysis

A First Look at Coding in Chemistry Tamas Bansagi, 2025-07-25 Do you have a project that requires some coding skills and don't know where to start Are you fitting a curve to your experimental data and need tools to help Trying coding to solve maths problems in chemistry for the first time This book will uncover all the help you need to apply these skills Focusing on developing basic coding skills in chemistry this concise introductory text takes a problem based approach and is organized in a workbook style for helping those new to programming Discussions on coding

are viewed from a chemistry perspective and embedded in solving problems familiar to most first year undergraduate chemistry students While primarily building programming and broadly applicable related skills some topics in data analysis and presentation uncertainties in measurements and areas of mathematics relevant to these and chemistry are also briefly surveyed After quick overviews of coding in chemistry and the MATLAB language and programming environment the basics of assigning variables and arrays are introduced Writing simple scripts and functions step by step through chemistry and environmental chemistry examples is then employed to present scientific data Coding skills are further developed in the next set of chapters while fitting curves to experimental data and handling measurement uncertainties Using coding to organise information in chemistry representing various types of transformations spatial effects and interactions is covered in the chapters on vectors and matrices Next user defined functions are utilized to predict the concentrations of chemicals during reactions before looking into developing code for enhanced workflows that allow computers to make decisions while executing programs In the final two chapters coding is extended to tackling common maths problems in chemistry including unit algebra rearranging expressions solving equations differentiation and integration

Catalytic Oxidation: Principles And Applications - A Course Of The Netherlands Institute For Catalysis Research (Niok) R A Sheldon,Rutger A Van Santen,1995-10-30 This book consists of lectures presented by international authorities in the field at a course on Oxidation Catalysis organized by the Dutch Research School in Catalysis at Rolduc in June 1994 The material covered spans the whole range of the subject from the fundamental principles of gas and liquid phase oxidations to reactor engineering for industrial processing The use of catalytic oxidation in both bulk and fine chemicals manufacture and the different types of catalysis heterogeneous gas phase homogeneous liquid phase and heterogeneous liquid phase are discussed In addition a few special topics such as electrocatalytic and high temperature oxidation are dealt with The book is intended for graduate students or industrial researchers who wish to acquaint themselves with the underlying principles of catalytic oxidations and the numerous applications of this important technology

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