

Name: _____

UNIT 4: Chemical Reactions Note Taking Document

Guidelines for making and writing observations:

1. Describe as many properties as you can.
2. Record detailed information that might help a chemist identify the substance.
3. Use descriptive words (such as thick, pink liquid) rather than imprecise language. For example, **DO NOT SAY**, "light-colored stuff."
4. Avoid opinions (such as looks gross)

Vocabulary:

- _____: something that can be _____ about a substance such as its color, smell, or boiling point.

1. Chemists rely on some properties that can only be observed under _____ such as _____ and _____ points.
2. Different substances have _____ melting and boiling points. Therefore, it is a _____ that could be used to help _____ a substance.
3. _____ is the temperature at which a solid will begin to melt and turn into a liquid.
4. _____ is the temperature at which a liquid will begin to boil and turn into vapor.

PHYSICAL CHANGE

1. Physical changes may change the state of matter but the substance _____ change.
 - Atoms do _____ change.
 - No new _____ form.
 - The state of matter or shape may _____
 - Can be _____

CHEMICAL CHANGE

1. Chemical changes create an entirely _____ substance. Referred to as _____
 - Brings about something _____
 - The chemical bonds of the molecules _____ and they reform in a different manner.
 - Chemical changes _____ be undone.

Key Concept

1. Different substances have _____ properties.

Note Taking Study Guide Chemical Reactions

Clifford Lane



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