

[4319]-1

F.Y.B.Sc. (Biotechnology)

CHEMISTRY

Bb - 101 : Fundamentals of Chemistry
(2008 Pattern)

Time : 3 Hours]

[Max. Marks : 80

Instructions to the candidates:

- 1) All questions are compulsory.
- 2) Draw neat and labelled diagrams wherever necessary.
- 3) Figures to the right indicates full marks.
- 4) Use of logarithmic table and calculator is allowed.

Q1) Answer the following :

[16]

- a) A sample of Helium has volume of 520 mL at 100°C. Calculate the temperature at which the volume will become 260 mL. Assume that pressure remains constant.
- b) Distinguish between order and molecularity.
- c) What is meant by abnormal molecular weight.
- d) Define the term component. How many components are present in the following system :



- e) What is effect of dilution of equivalent conductance of strong electrolyte.
- f) How will you determine the solubility product of sparingly soluble salt by emf method?
- g) What are distereomers?
- h) Distinguish between sigma (σ) and pi (Π) bond.

Q2) Attempt any FOUR of the following :

[16]

- a) State and explain Dalton's law of partial pressure.
- b) Show that half life period of first order reaction is independent of initial concentration of reactants.
- c) What are colligative properties? Explain any one colligative property.
- d) Describe KI - H₂O system on the basis of phase rule.
- e) Discuss Debye - Huckel theory of activity coefficients of strong electrolyte.
- f) Explain the factors affecting Vander Waal's forces.

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Sqa Specimen Paper 2014 Past Paper National 5 Chemistry and Hodder Gibs Sqa (Hodder Ed),2014-10-31

Chemistry 370 Christopher Nichols,2017-07-03 *Higher Level Thinking Questions* Angela Manzi,Miguel (CRT)

Kagan,Michael Michels,2009-12-17 Your secondary students will explore chemistry topics and themes like never before with this giant collection of ready to use chemistry questions You ll find questions for eighteen chemistry topics and themes Acids and Bases Atomic Structure Biochemistry Bonding Chemical Industry Chemical Reactions Electron Configurations Environmental Chemistry and Pollution Gases Hydrocarbons and Petroleum Metals Methods and Tools Nuclear Chemistry and Alternative Energy Sources Periodic Table Solutions States of Matter Stoichiometry and Water Nurture the development of your students thinking skills with hundreds of questions such as How can gas pressure affect the health of someone who goes scuba diving Why does maple sugar flow from a tree in winter Promote thinking and active engagement in your chemistry class with this great book **Questions in Chemistry** A. L. H. SMITH,1972 **Questions on Chemistry**

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Table of Contents Question Set Fybsc 2013 Chemistry

1. Understanding the eBook Question Set Fybsc 2013 Chemistry
 - The Rise of Digital Reading Question Set Fybsc 2013 Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Question Set Fybsc 2013 Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Question Set Fybsc 2013 Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Question Set Fybsc 2013 Chemistry
 - Personalized Recommendations
 - Question Set Fybsc 2013 Chemistry User Reviews and Ratings
 - Question Set Fybsc 2013 Chemistry and Bestseller Lists
5. Accessing Question Set Fybsc 2013 Chemistry Free and Paid eBooks
 - Question Set Fybsc 2013 Chemistry Public Domain eBooks
 - Question Set Fybsc 2013 Chemistry eBook Subscription Services
 - Question Set Fybsc 2013 Chemistry Budget-Friendly Options
6. Navigating Question Set Fybsc 2013 Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Question Set Fybsc 2013 Chemistry Compatibility with Devices
 - Question Set Fybsc 2013 Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Question Set Fybsc 2013 Chemistry
 - Highlighting and Note-Taking Question Set Fybsc 2013 Chemistry
 - Interactive Elements Question Set Fybsc 2013 Chemistry
8. Staying Engaged with Question Set Fybsc 2013 Chemistry

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Question Set Fybsc 2013 Chemistry
- 9. Balancing eBooks and Physical Books Question Set Fybsc 2013 Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Question Set Fybsc 2013 Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Question Set Fybsc 2013 Chemistry
 - Setting Reading Goals Question Set Fybsc 2013 Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Question Set Fybsc 2013 Chemistry
 - Fact-Checking eBook Content of Question Set Fybsc 2013 Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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