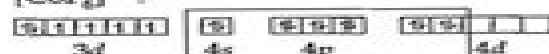


Now electronic configuration Co^{3+} :



Thus, outer orbital or high spin complex of $[\text{CoF}_6]^{3-}$:



(Six pairs of electrons from six fluoride ions.)

- (b) Dibromide bis(ethane-1,2-diamine) Cobalt(III) is the IUPAC name of $[\text{CoBr}_2(\text{en})_2]^+$ complex.

- (c) The number of ions produced from the complex:

$[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$ in solution is three



7. (a) Differentiate between the following:

11 3 3 3 3

- (i) Adsorption and Absorption
- (ii) Lyophobic Sol and Lyophilic Sol
- (iii) Multimolecular Colloid and Macromolecular colloid

012

- (b) (i) Define the following terms:

- #### ④ Zeta Potential

- (d) **Coagulation**

- (II) Why a negatively charged sol is obtained when AgNO_3 solution is added to KI solution? [3]

8. Define transition metals. Why Zn, Cd and Hg are not called transition metals? How is the variability in oxidation states of transition metals different from that of s-block elements?

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Ans. Transition metals are the metals that have partially filled *d*-orbitals and having the ability to form stable cations.

Zn, Cd, Hg are not regarded as transition metals because they have completely filled d -subshell with outer electronic configuration $(n-1)d^{10}ns^2$

The variability in oxidation states of transition metals is due to the incomplete filling of *d*-orbitals in such a way, that their oxidation states differ from each other by unity (i.e., Fe^{2+} and Fe^{3+}). While, in case of *p*-block elements, the oxidation state differ by two units (i.e., +3 and +5).

9. (a) What happens when: [1 × 3 = 3]

11. $3 = 31$

- (i) Propanone is treated with CH_3MgBr and then hydrolysed?

- (ii) Ethanal is treated with excess ethanol and acid?

- (iii) Methanal undergoes Cannizzaro reaction?

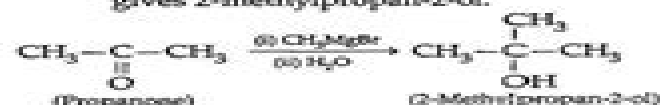
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- (b) Write the main product in the following reactions: [1 × 3 = 3]

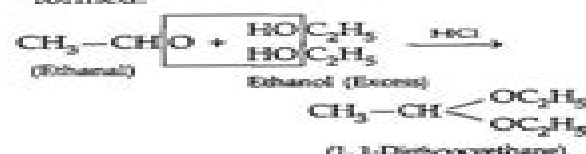
- $$(i) \quad 2 \text{CH}_3\text{COO}^- + (\text{CH}_3)_3\text{C}^+ \longrightarrow$$



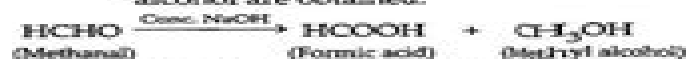
Ans. (a) (i) When propanone is treated with CH_3MgBr followed by hydrolysis it gives 2-methylpropan-2-ol.



- (ii) When ethanal reacts with excess of ethanol in presence of trace amount of HCl then, 1, 1-Dithoxyethane will be formed.



- (iii) When methanal undergoes Cannizzaro reaction then, formic acid and methyl alcohol are obtained.



- $$(b) \quad (i) \quad 2 \text{CH}_3\text{COCl} + (\text{CH}_3)_2\text{Cd} \longrightarrow$$



Thus, 2-Propanone is the main product.



Thus, propane is the main product.



Thus, cyclohexane is the main product

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M Lipman



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