

## Physical Science Chapter 4

1000 points

Thermography

Exotherm

Endotherm

Heat

Freeze

Boiling

Melting

Condensing

Evaporation/boiling point

Specific Heat

The specific heat of soil is  $0.20 \text{ kcal/kg}^\circ\text{C}$  and the specific heat of water is  $1.00 \text{ kcal/kg}^\circ\text{C}$ . This means that if 1 kg of soil and 1 kg of water each receives 1 kcal of energy, ideally,

**The soil will be  $5^\circ\text{C}$  warmer than the water.**

The specific heat of water is  $1.00 \text{ cal/g}^\circ\text{C}$  and the specific heat of air is  $0.167 \text{ cal/g}^\circ\text{C}$ . The same amount of energy applied to equal

# Physical Science Chapter 4 Test

**N Noddings**



## **Physical Science Chapter 4 Test:**

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