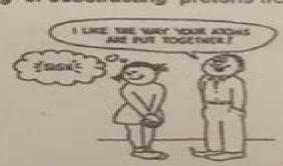
Three fundamental par	ticles of the atom a	re the		, and
At th	a center of each at	om lies the atomic		which
consists of				
number of				
same number of				
	oms that have the			but a different
number of	An isotope	is identified by its	atomic mass	number, which
is the total number of	and ,	in th	e nucleus. A ca	irbon isotope
that has 6	and 6	is identif	led as carbon-1:	2, where 12 is
the atomic mass num	carbon-14.	tope having 6	and 8	3
1. Complete the follow	ving table:			
	ving table:	Number of		1
	electrons	Number of Protons	Neutrons	1

Isotope	Electrons	Number of Protons	Neutrons
Hydrogen-1	1		
Chlorine-36		17	
Nitrogen-14			7
Potassium-40	19		
Arsenio-75		33	
Gold-197			118

- 2. Which results in a more valuable product adding or substracting protons from gold nuclei?
- 3. Which has more mass, a helium atom or a neon atom?
- 4. Which has a greater number of atoms, a gram of helium or a gram of neon?



# **Physical Science Chapter 12 Answers**

F Rizvi

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