2006

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Your Roll No.....

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B.Sc. (Hons.)/III/NS

F

CHEMISTRY—Paper XVII

(Physical Chemistry—III)

Time : 3 Hours

Maximum Marks: 38

(Write your Roll No. on the top immediately on receipt of this question paper.)

Attempt six questions in all.

Q. No. 1 is compulsory. At least one question is to be attempted from each Section.

- Explain briefly any four of the following: 4×2=8
  - (a) Maximum order of a reaction is restricted to three only.
  - (b) A photochemical reaction with quantum efficiency of one shows no temperature dependence.
  - (c) Adsorption of gases on the surface of adsorbent is not monolayer at high pressure and low temperature.
  - (d) Why can we not use a voltmeter for determining the electromotive force of a galvanic cell?
  - (e) A first order reaction never completes.

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**JR Anderson** 

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