

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.100 M volume 25.00 mL acid base 0.100 M

b. 0.100 M volume 25.00 mL acid base 0.100 M

c. 0.100 M pH 4.75 compound that releases acetate
0.100 M acid base 0.100 M acid

Acetic Acid: 1st, 2nd, 3rd, 4th, 5th, 6th, 7th, 8th, 9th, 10th

1.00 2.00 3.00 4.00 5.00 6.00 7.00 8.00 9.00 10.00

Species	Acid (pH, Base pH, or Salt pH)	Acids with M_1 ($M_1 = M_2$)	Conductivity (C or hC)
CH_3COOH	<u>4.75</u>	<u>1</u>	<u>1</u>
CH_3COO^-	<u>9.25</u>	<u>1</u>	<u>1</u>
$NaCH_3COO$	<u>9.25</u>	<u>1</u>	<u>1</u>
$NaCH_3COO^-$	<u>9.25</u>	<u>1</u>	<u>1</u>

Note Taking Guide 1101 Chemistry Answers

S Ben Porath



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