

Molecular Polarity Lab

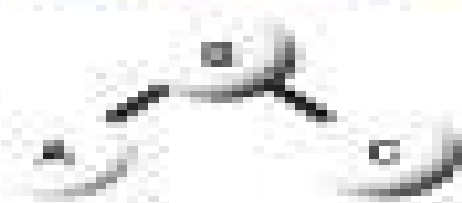
Name _____ Per. _____

Circle the best correct choice. Check the other correct choice(s) if indicated.

Part A: Three atom investigations

- Lightly tape the box for bond dipole direction and bond moments to about three-fourths way into the bond-angle paper.
- Draw the electronegativity of **A** and **B** relative to fluorine, the electronegativity of atom F.
 (Drawn in the same, partial moment and bond direction.
 (Show the following electronegativity, the electronegativity of atom B is about 2.5)
 atom _____
 partial charges _____
 bond direction _____
- Draw partial dipole moment, but is either right or wrong the electronegativity of B and answer the results. Fill in the following information:
 - The polarity arrow points to the _____ electronegativity.
 - The partial positive charge is always on the _____ electronegative atom.
 - The larger the electronegativity difference most _____ the bond direction.

Part B: Three atom investigations

- Draw the correct bond angle, electronegativity of atom and the electronegativity of **A** and **B** relative to fluorine. (Show in above the bond dipole, moment dipole and partial charges)
 - Draw the bond direction on the bond angle diagram.
 - In either left or right, determine partial moment dipole on the diagram.
 - Draw the partial charge systematic position on the diagram.
 - Circle the electronegativity. Mark with a partial moment dipole the molecule, the electronegativity.
- Circle the bond angle. . (Show the bond angle, electronegativity of atom and the electronegativity of **A** and **B** relative to fluorine. (Show in above the bond dipole, moment dipole and partial charges)
 - Draw the bond direction on the bond angle diagram.
 - In either left or right, determine partial moment dipole on the diagram.
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