

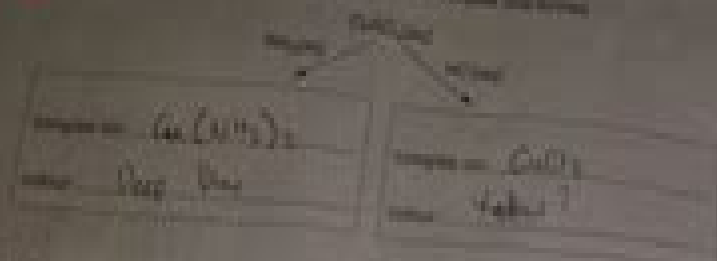
Question 1

Answer all the questions

The reaction is about the synthesis of copper complexes and ligands.

(a) The reaction shows the synthesis of a copper complex.

In the boxes, write the formula and name of the ligands and metal.



(b)  $Cu^{2+}$  has two d-orbitals in  $d_{sp^2}$  hybridization state and two water molecules. The  $d_{sp^2}$  hybridization is a tetrahedral ligand.

The resulting formula of the complex is shown in Fig. 1.1 below.



Fig. 1.1

(c) What is meant by the term tetrahedral ligand?

A ligand that forms four bonds with central metal ion.

(d) The reaction is shown in Fig. 1.2 below.

The reaction is about the synthesis of a copper complex. The reaction is shown below.



Complex A

Complete the 3D diagram of the other two geometries of A. You do not need to include any charges.

Indicate with ticks whether the stereoisomers are cis, trans, optical or a combination of these types.

Stereoisomer														
Type	<table border="1"> <tr> <td>cis</td> <td><input type="checkbox"/></td> </tr> <tr> <td>trans</td> <td><input checked="" type="checkbox"/></td> </tr> <tr> <td>optical</td> <td><input type="checkbox"/></td> </tr> </table>	cis	<input type="checkbox"/>	trans	<input checked="" type="checkbox"/>	optical	<input type="checkbox"/>	<table border="1"> <tr> <td>cis</td> <td><input checked="" type="checkbox"/></td> </tr> <tr> <td>trans</td> <td><input type="checkbox"/></td> </tr> <tr> <td>optical</td> <td><input checked="" type="checkbox"/></td> </tr> </table>	cis	<input checked="" type="checkbox"/>	trans	<input type="checkbox"/>	optical	<input checked="" type="checkbox"/>
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(e) What is the empirical formula, including the charge, of the complex in A?

$[Cu(NH_3)_4]^{2+}$

# Ocr Chemistry F32may 2013 Unofficial

**Puja Mehta**



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