

# Salters OCR B chemistry EL not mine

2 bond pairs, 0 lone pairs - ANSLinear

180°

e.g. BeCl<sub>2</sub>

3 bond pairs, 0 lone pairs - ANSTrigonal Planar

120°

e.g. BF<sub>3</sub>

2 bond pairs, 1 lone pair - ANSBent

118°

e.g. NO<sub>2</sub>

4 bond pairs, 0 lone pairs - ANSTetrahedral

109.5°

e.g. CH<sub>4</sub>

3 bond pairs, 1 lone pair - ANSPyramid

107°

e.g. NH<sub>3</sub>

2 bond pairs, 2 lone pairs - ANSBent

104.5°

e.g. H<sub>2</sub>O

5 bond pairs, 0 lone pairs - ANSTrigonal Bipyramid

120°, 90°

e.g. PCl<sub>5</sub>

6 bond pairs, 0 lone pairs - ANSOctahedral

90°

e.g. SF<sub>6</sub>

In 1803 John Dalton said... - ANSThat all atoms are SPHERES and all elements are made from different spheres

Plum pudding model - ANSIn 1897 J.J. Thompson discovered the electron and said there were negative electrons "plums" in a positive "pudding"

Gold Leaf Experiment - ANS1909, Ernest Rutherford.

Fired positive alpha particles at gold leaf. Most went through to other side, some deflected back.

He concluded that the atom had a very small positively charged nucleus and a negative cloud of empty space.

# Ocr Salters B Chemistry Skill 3 All Specimens

**Yijin Wang**



### **Ocr Salters B Chemistry Skill 3 All Specimens:**

**OCR(B) A2 Chemistry (Salters) Student Unit Guide: Unit F334 Chemistry of Materials** Frank Harriss, 2009-06-26

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